Toshihiko Ueki [a], Matsumi Doe [a], Rika Tanaka [b], Yoshiki Morimoto [a], Kazuo Yoshihara [c] and Takamasa Kinoshita* [a]<br>Department of Chemistry, Graduate School of Science, Osaka City University, Sumiyoshi-ku, Osaka 558-8585, Japan<br>Department of Applied Chemistry, Faculty of Engineering, Osaka City University<br>Suntory Institute for Bioorganic Research, Shimamoto-cho, Osaka 618-0024, Japan<br>Received May 22, 2000

Hyperolactones A (1) and C (3) have been synthesized starting from (S)-malic acid by a straightforward route. The unique spirolactone skeleton was efficiently constructed by one-pot reaction as a key step. The absolute stereochemistry of hyperolactones was unambiguously established by this synthesis.
J. Heterocyclic Chem., 38, 165 (2001).
(-)-Hyperolactone A (1) was isolated from the leaves and stems of the Chinese plant Hypericum chinense L. by M. Tada and co-workers in 1989 [1]. Later, two structural relatives of 1, (+)-hyperolactone B (2) and (-)-hyperolactone $\mathrm{C}(\mathbf{3})$, were also isolated from the same plant as minor components in 1995 [2]. The gross structure of $\mathbf{1}$ and its relative stereochemistry were determined by means of extensive spectroscopic studies and chemical transformation, and was finally confirmed by X-ray crystallographic analysis. These compounds possess a common spirolac-


Hyperolactone A (1)


Hyperolactone B (2)


Hyperolactone C (3)

Figure 1
tone skeleton (anti-orientation at the stereogenic center C-5 in $\mathbf{2}$ as compared with $\mathbf{1}$ and $\mathbf{3}$ ), and some novel characteristics of the structure are (1) a highly substituted 1,7dioxaspiro $[4,4]$ non-2-ene-4,6-dione skeleton; (2) a strained $2,2,5$-trisubstituted $3(2 H)$ furanone ring system. Owing to their structural uniqueness, compound 1 and structurally related natural products are good targets for total synthesis [3]. We now describe herein the synthesis of $\mathbf{1}$ and $\mathbf{3}$ in a stereoselective fashion, which enables us to establish their absolute stereochemistry unambiguously.

## Synthesis of the Side Chain Units 8 and 14.

The requisite aldehyde $\mathbf{8}$ was synthesized as shown in Scheme 1. The $\beta$-keto ester $\mathbf{6}$ was obtained by the Collins oxidation of the $\beta$-hydroxy ester 5 , which was prepared from ethyl bromoacetate and benzaldehyde $\mathbf{4}$ by the Reformatsky reaction [4]. The keto ester $\mathbf{6}$ was heated with ethylene glycol containing a small amount of $p$-toluenesulfonic acid in benzene to give the ketal 7 in $73 \%$ yield. Diisobutylaluminium hydride reduction of $\mathbf{7}$ afforded the aldehyde $\mathbf{8}$ in $78 \%$ yield.

Scheme 1


Reagents and conditions: i, $\mathrm{Zn}, \mathrm{BrCH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$, benzene, reflux, $85 \%$; ii, $\mathrm{CrO}_{3}$, pyridine, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{rt}, 82 \%$; iii, ethyleneglycol, $p$-TsOH, benzene, reflux, $73 \%$; iv, diisobutylaluminium hydride, toluene, $-78^{\circ}, 78 \%$.

Scheme 2



PMB=p-methoxybenzyl
Reagents and conditions: i, ref. 5; ii, $\mathrm{LiAlH}_{4}$, tetrahydrofuran, reflux, $83 \%$; iii, $\left[\mathrm{PhCH}_{2} \mathrm{NEt}_{3}\right]^{+} \mathrm{Cl}^{-}, \mathrm{MsCl}, \mathrm{NaOH}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, $62 \%$; iv, 1,3-dithiane, n-BuLi, tetrahydrofuran, $-40^{\circ}$ to $0^{\circ}, 79 \%$; v, NaH, p-methoxybenzyl chloride, dimethylformamide, $84 \%$. vi, $\mathrm{HgCl}_{2}, \mathrm{HgO}, \mathrm{CH}_{3} \mathrm{CN}$, reflux, $84 \%$.

Since the X-ray crystallographic analysis of 1 suggests (2'S)-configuration, a synthetic plan for the requisite (4S)aldehyde 14 as a coupling partner is shown in Scheme 2. Reduction of ( $2 S, 3 S$ )-acetoxy-3-methylvaleric acid 9 [5], prepared from $(2 S, 3 S)$-isoleucine, with lithium aluminium hydride afforded the diol 10 in $83 \%$ yield, which was converted [6] into the volatile epoxide 11 in $62 \%$ yield. The lithiation of 1,3 -dithiane with $n$-butyllithium and subsequent addition to the epoxide $\mathbf{1 1}$ gave the dithiane $\mathbf{1 2}$ in $79 \%$ yield. p-Methoxybenzyl protection of the hydroxyl group in 12 and subsequent oxidative cleavage of the dithiane afforded the desired aldehyde 14.
Synthesis of 3 (Scheme 3).
The aldol reaction of the lithium enolate of $(3 S, 4 S)$-3-methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone $\mathbf{1 5}$ (Scheme 5, see experimental section) [3b] with the aldehyde 8 afforded 16 as an inseparable mixture of four diastereomers in $83 \%$ yield. Although a multitude of isomers is formed, in the light of the subsequent steps this factor is of minor significance. Successive treatment of $\mathbf{1 6}$ with the Jones reagent provided 17 in $92 \%$ yield ( $a: b=2: 3$ ) as a separable mixture.

The stereochemistries of 17 were confirmed by ${ }^{1} \mathrm{H}$ NMR analysis and NOE experiments of 17a and 17b
(Figure 2). Both in 17a and 17b, a NOE was observed between the 4-methyl group ( $4-\mathrm{CH}_{3}$ ) and a higher-field hydrogen (5-Ha) of the 5-methylene, but not observed between the $4-\mathrm{CH}_{3}$ and a lower-field hydrogen $(5-\mathrm{Hb})$. In 17a, a small NOE was observed between the $4-\mathrm{CH}_{3}$ and a methylene hydrogen of the methoxymethyloxy group, whereas a NOE was not observed in 17b. By the anisotropic effect of the methoxymethyloxy group, the signals of $4-\mathrm{CH}_{3}$ and 5-Ha for $\mathbf{1 7 a}$ appeared in the lower field than the corresponding signals for $\mathbf{1 7 b}$, and $5-\mathrm{Hb}$ signal for 17b was observed in the lower field than that for $\mathbf{1 7 a}$. These results indicate that the methoxymethyloxy group and the $4-\mathrm{CH}_{3}$ in $\mathbf{1 7 a}$ were located in the syn relationship. From these considerations, it was found that the stereochemistry of $\mathbf{1 7 a}$ was $(3 S, 4 S)$ - configuration and $\mathbf{1 7 b}$ was its $3 R$-epimer.

Acid-catalyzed hydrolysis of 17a in boiling tetrahydrofuran containing $3 M$ hydrochloric acid furnished $\mathbf{3}$ (93\% yield) consequently via a subsequent ring closure and dehydration (Scheme 3). The compound $\mathbf{3}$ was found to be identical with the natural product [2] by comparing the mp , $[\alpha]_{D}, \mathrm{~ms},{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra with those reported, while the physical data of compound 3a derived from 17b were all different except the hrms.

Scheme 3


MOM = methoxymethyl
Reagents and conditions: i, lithium diisopropylamide, tetrahydrofuran, $-78^{\circ}, 83 \%$; ii, $\mathrm{CrO}_{3}, \mathrm{H}_{2} \mathrm{SO}_{4}$, acetone, $0^{\circ}, 92 \%$; iii, 3 M HCl , tetrahydrofuran, reflux, $\mathbf{3}$ ( $93 \%$ ); 3a(100\%).


Reagents and conditions: i, (a) lithium diisopropypamide, tetrahydrofuran, $-78^{\circ}, 84 \%$ (b) Dess-Martin periodinane, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 100 \%$; ii, dichlorodicyanobenzoquinone, $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{H}_{2} \mathrm{O}, 94 \%$; iii, $\mathrm{CrO}_{3}, \mathrm{H}_{2} \mathrm{SO}_{4}$, acetone, $-20^{\circ}, 66 \%$; iv, trimethylsilyl bromide, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-20^{\circ}, 98 \%$.


17a


17b

Figure 2
$\mathrm{ms},{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra with those reported [1]. The $\beta$-methoxymethyloxy isomer 20b was treated with bromotrimethylsilane to afford a $5 R$-isomer 1a as a single product in $95 \%$ yield.

Thus, we have completed the total synthesis of $\mathbf{1}$ and $\mathbf{3}$. This synthesis discloses that the stereochemistry should be depicted as ( 2 'S,5S,9S)-2-(2'-butyl)-9-ethenyl-9-methyl-1,7-dioxaspiro-[4,4]non-2-ene-4,6-dione for (-)-hyperolactone A and (5S,9S)configuration for (-)-hyperolactone C , respectively.

Scheme 5
Preparation of $\mathbf{1 5}$


Reagents and conditions: i, lithium diisopropylamide, $\mathrm{PhCH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{I}$, tetrahydrofuran, $-78^{\circ}, 67 \%$; ii, KOH ,
MeOH- $\mathrm{H}_{2} \mathrm{O}, \mathrm{rt}, 97 \%$; iii, $\mathrm{LiB}\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{H}$, tetrahydrofuran, rt, $74 \%$; iv, $\mathrm{CH}_{3} \mathrm{OCH}_{2} \mathrm{Cl}, i-\mathrm{Pr}_{2} \mathrm{EtN}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, reflux, $93 \%$; v, $10 \%$
Pd-C, $\mathrm{H}_{2}, \mathrm{EtOH}, 96 \%$; vi, $o-\mathrm{O}_{2} \mathrm{NC}_{6} \mathrm{H}_{4} \mathrm{SeCN}, \mathrm{Bu}_{3} \mathrm{P}$, tetrahydrofuran, $\mathrm{rt}, 88 \%$; vii, $30 \% \mathrm{H}_{2} \mathrm{O}_{2}$, tetrahydrofuran, $\mathrm{rt}, 76 \%$.

Synthesis of 1 (Scheme 4).
A similar coupling reaction of $\mathbf{1 5}$ with the aldehyde 14 gave an inseparable mixture of four diastereomers of the hydroxyl compound in $84 \%$ yield, which on oxidation with Dess-Martin periodinane provided 18 in quantitative yield ( $a: b=1: 1$ ). The two diastereoisomers 18a and $\mathbf{1 8 b}$ were separated by chromatography. By similar ${ }^{1} \mathrm{H}$ NMR analysis and NOE experiments of 18, a $2.2 \%$ enhancement was observed in the signal for the 4-methyl group upon irradiation of the methylene proton of the methoxymethyloxy group in 18a, thereby confirming that the desired stereochemistry had been established. Deprotection of the $p$-methoxybenzyl group in the lactone 18a was accomplished using dichlorodicyanobenzoquinone in dichloromethane to give the $\beta$-hydroxy ketone 19a in $94 \%$ yield. The Jones oxidation of 19a under mild condition resulted in enol formation of the diketone 20a, the structure of which was elucidated by NMR.

An acid-catalyzed reaction of 20a with boiling tetrahydrofuran in the presence of $3 M$ hydrochloric acid gave a mixture of diastereomers of $\mathbf{1}$ and its $2^{\prime}$-isomer $\mathbf{1 b}$ ( $1: 1$ by NMR) in $70 \%$ yield. Such drastic conditions indeed would be inappropriate for the delicate ketone. Finally and much to our relief, deprotection of 20a was achieved using bromotrimethylsilane in dichloromethane at $-20^{\circ}$ to give the spirolactone $\mathbf{1}$ as a single product in $98 \%$ yield. The synthetic 1 was found to be identical with the natural product by comparing the $\mathrm{mp},[\alpha]_{\mathrm{D}}$,

The advantages of this strategy are the short and efficient steps in one-pot direct construction of spirolactone by acidcatalysis via deprotection, cyclization and elimination.

## EXPERIMENTAL

Melting points were measured with a Yanagimoto MP apparatus and are uncorrected. Specific rotations were measured with a JASCO Model DIP-370 polarimeter. Ir spectra were obtained on a JASCO A-102 infrared spectrophotometer. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr} \mathrm{spectra}$ were recorded on a JEOL LA-300 $(300 \mathrm{MHz})$ in deuterio chloroform solution, unless stated otherwise. Chemical shifts (ppm) are given downfield of tetramethylsilane. Mass spectra were determined on a JEOL AX-500 spectrometer. Elemental analyses were performed with a JMS AX-500 elemental analyzer by the staff at our Instrumental Measurement Center. Thin-layer chromatography (TLC) was performed with a glass plate coated Kieselgel $60 \mathrm{GF}_{254}$ (Merck). Column chromatography was carried out on silica gel 60 (Merck No. 7734; 63-200 $\mu \mathrm{m}$ ). Solvents were dried (drying agent in parenthesis) and distilled prior to use: tetrahydrofuran and diethyl ether (sodium/benzophenone ketyl), benzene and dichloromethane (phosphorous pentoxide), and dimethylsulfoxide (calcium hydride). Organic solutions were dried over anhydrous sodium sulfate.
Ethyl 3-hydroxy-3-phenylpropanoate (5).
This compound was prepared from benzaldehyde and ethyl bromoacetate by the Reformatsky reaction in $85 \%$ yield as a colorless oil; bp $120-123^{\circ} / 2 \mathrm{~mm} \mathrm{Hg}$; (lit [4] $105^{\circ} / 0.2 \mathrm{~mm} \mathrm{Hg}$ ).

Ethyl 3-Phenyl-3-oxopropanoate (6).
This compound was prepared by the Collins oxidation [7] of $\mathbf{5}$ in $82 \%$ yield as a colorless oil: bp $115 \% 1 \mathrm{~mm} \mathrm{Hg}$; (lit [7] $95 \% / 0.2$ $\mathrm{mm} \mathrm{Hg})$; ir (film) $1725,1670,1620 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ : $\delta 1.25(\mathrm{t}, 3 \mathrm{H}$, $\mathrm{J}=7.1 \mathrm{~Hz}), 3.99(\mathrm{~s}, 2 \mathrm{H}), 4.21(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 7.37-7.96(\mathrm{~m}$, $5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 14.1$ (q), 46.0 (t), 61.5 (t), 126.0 (d), 128.5 (d), 128.8 (d), 133.7 (s), 167.5 (s), 192.5 (s). hrms: Calcd. for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}:\left(\mathrm{M}^{+}\right) 192.0786$. Found: m/z 192.0775.

## Ethyl 3,3-Ethylenedioxy-3-phenylpropanoate (7).

A solution of $\mathbf{6}$ ( 10 mmoles ), ethyleneglycol ( 20 mmoles ) and $p$ toluenesulfonic acid ( 20 mg ) in benzene ( 50 ml ) was refluxed for 15 hours. The cooled mixture was poured onto saturated aqueous sodium bicarbonate ( 50 ml ), and the organic layer was washed with water and dried. The solvent was removed and the residue was distilled to give $\mathbf{7}$ in $73 \%$ yield as a colorless oil; bp $121^{\circ} / 2 \mathrm{~mm} \mathrm{Hg}$; ir (film) $1720 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.15(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 2.96(\mathrm{~s}, 2 \mathrm{H})$, $3.80-3.84(\mathrm{~m}, 2 \mathrm{H}), 4.03-4.11(\mathrm{~m}, 2 \mathrm{H}), 4.07(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 7.27-$ 7.52 (m, 5H); ${ }^{13} \mathrm{C}$ nmr: $\delta 14.0$ (q), 45.8 (t), 60.4 (t), 64.8 (t), 107.8 (s), 125.6 (d), 128.1 (d), 128.2 (d), 141.4 (s), 168.5 (s). hrms: Calcd. for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{O}_{4}:(\mathrm{M}+\mathrm{H})^{+}$237.1129. Found: m/z 237.1127.

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{4}$ : C, 66.09; H, 6.83. Found: C, 66.00; H, 6.88.

## 3,3-Ethylenedioxy-3-phenylpropanal (8).

To a solution of $\mathbf{7}(1.0 \mathrm{mmol})$ in toluene $(10 \mathrm{ml})$ was added diisobutylaluminium hydride ( 1.5 M in toluene, 1.01 mmoles ) at $-78^{\circ}$. After stirring for 10 minutes the reaction mixture was quenched by addition of water. The precipitate was removed by filtration through celite. The filtrate was washed with water, dried and distilled to give $\mathbf{8}$ in $78 \%$ yield as white crystals; $\mathrm{mp} 87^{\circ}$; ir (nujol) $1710 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}: \delta 2.90(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=2.9 \mathrm{~Hz}), 3.78-3.90(\mathrm{~m}, 2 \mathrm{H})$, 4.03-4.14 (m, 2H), 7.29-7.50 (m, 5H), $9.76(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=2.9 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 52.7$ (t), 64.7 (t), 108.2 (s), 125.3 (d), 128.5 (d x 2), 141.2 ( s$)$, 199.8 (d). MS (EI) m/z 192 (M+, 15\%), 149 (90), 115 (100).

Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}$ : C, 68.74; H, 6.29. Found: C, 68.48; H, 6.33.

## (2S,3S)-3-Methyl-1,2-pentanediol (10).

To a stirred suspension of lithium aluminium hydride ( 4.36 g , 115 mmoles ) in tetrahydrofuran ( 60 ml ) was added a solution of (2S,3S)-2-acetoxy-3-methylvaleric acid 9 [5] ( $5.0 \mathrm{~g}, 28.7$ mmoles) in tetrahydrofuran ( 30 ml ) at $0^{\circ}$. After refluxing for 5 hours a small amount of water was added dropwise to decompose excess lithium aluminium hydride, and the precipitate was filtered off through celite and washed with ethyl acetate. The combined filtrates were dried and concentrated to give $\mathbf{1 0}$ as a colorless oil ( $2.8 \mathrm{~g}, 83 \%$ ); bp $87^{\circ} / 9 \mathrm{~mm} \mathrm{Hg} ;[\alpha]_{\mathrm{D}}^{24}+5.9$ (c 1.13, methanol); ir (film) $3400 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.87(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.7 \mathrm{~Hz}), 0.91(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.04-$ $1.24(\mathrm{~m}, 1 \mathrm{H}), 1.50-1.64(\mathrm{~m}, 2 \mathrm{H}), 3.23(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.49(\mathrm{dd}, 2 \mathrm{H}, \mathrm{J}=$ $8.2 \mathrm{~Hz}, \mathrm{~J}=17.1 \mathrm{~Hz}), 3.64-3.70(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 11.2(\mathrm{q}), 14.6$ (q), 25.0 (t), 37.6 (d), 64.2 (t), 76.0 (d).

## ( $2 S, 3 S$ )-3-Methyl-1,2-epoxypentane (11).

A mixture of $\mathbf{1 0}$ ( $2.5 \mathrm{~g}, 21.2$ mmoles), benzyltriethylammonium chloride [6] ( $139 \mathrm{mg}, 3 \mathrm{~mol} \%$ ) and $20 \%$ aqueous sodium hydroxide ( 25 ml ) in 25 ml of dichloromethane was heated under reflux, and a solution of mesyl chloride ( $3.6 \mathrm{~g}, 31.8$ mmoles) in 40 ml of dichloromethane was added in portion. Heating and stirring was continued until complete consumption of the starting diol. The mixture was cooled to room temperature, and was poured
into 25 ml of water. The organic layer was dried and distilled to give $11(1.3 \mathrm{~g}, 62 \%)$ as a colorless liquid; bp $95^{\circ} / 760 \mathrm{~mm} \mathrm{Hg}$; $[\alpha]_{\mathrm{D}}^{21}+3.8$ (c 2.10, ethanol); ${ }^{1} \mathrm{H}$ nmr: $\delta 0.92(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.7 \mathrm{~Hz})$, $0.96(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.15-1.66(\mathrm{~m}, 3 \mathrm{H}), 2.46-2.56(\mathrm{~m}, 1 \mathrm{H})$, 2.64-2.78 (m, 2H); ${ }^{13} \mathrm{C} \mathrm{nmr:} \delta 11.2$ (q), 15.1 (q), 27.2 (t), 37.6 (d), 45.6 (t), 56.9 (d).
( $2 R, 3 S$ )-2-(2-Hydroxy-3-methylpentyl)-1,3-dithiane (12).
To a solution of 1,3 -dithiane ( $3.01 \mathrm{~g}, 25.1 \mathrm{mmoles}$ ) in tetrahydrofuran ( 40 ml ) was added $n$-butyllithium ( 1.6 M in hexane, $14.2 \mathrm{ml}, 22.9 \mathrm{mmols}$ ) at $-40^{\circ}$ under nitrogen and the mixture was stirred for 2 hours. A solution of $\mathbf{1 1}(2.09 \mathrm{~g}, 20.9 \mathrm{mmoles})$ in tetrahydrofuran ( 15 ml ) was added to the reaction mixture, and stirred for additional 5 hours at $-40^{\circ}$. After standing overnight at $0^{\circ}$, the reaction mixture was quenched with saturated aqueous ammonium chloride, and extracted with ether ( $4 \times 100 \mathrm{ml}$ ). The combined organic layer was dried and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 5:1) to give $12(3.63 \mathrm{~g}, 79 \%)$ as a pale yellow oil; $[\alpha]_{\mathrm{D}}^{20}+28.3$ (c 0.91, ethanol); ir (film) $3350 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.89$ (d, 3H, J $=6.8$ $\mathrm{Hz}), 0.92(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.12-1.21(\mathrm{~m}, 1 \mathrm{H}), 1.44-1.53(\mathrm{~m}$, $2 \mathrm{H}), 1.80-1.96(\mathrm{~m}, 3 \mathrm{H}), 2.02-2.19(\mathrm{~m}, 1 \mathrm{H}), 2.84-2.94(\mathrm{~m}, 4 \mathrm{H})$, 3.78-3.90 (m, 1H), $4.28(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=5.5 \mathrm{~Hz}, \mathrm{~J}=8.3 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 11.6$ (q), 14.4 (q), 24.8 (t), 26.0 (t), 30.0 (t), 30.4 (t), 38.8 (d), 40.7 (t), 44.7 (d), 72.0 (d). MS (EI) m/z 220 ( $\mathrm{M}^{+}, 60 \%$ ), 145 (100), 119 (75); hrms: Calcd. for $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{OS}_{2}:(\mathrm{M})^{+} 220.0956$. Found: m/z 220.0982.
(2R,3S)-2-[2-(4-Methoxybenzyloxy)-3-methylpentyl]-1,3-dithiane (13).

To a suspension of sodium hydride ( $158 \mathrm{mg}, 3.96 \mathrm{mmoles}$ ) in dimethylformamide ( 15 ml ) was added $12(581 \mathrm{mg}, 2.64$ mmoles) in dimethylformamide ( 3 ml ) at room temperature, and the mixture was stirred for 1 hour. To the mixture was added p-methoxybenzyl chloride ( $1.07 \mathrm{ml}, 7.90$ mmoles), and the mixture was stirred for 21 hours at $50^{\circ}$. The reaction mixture was quenched with water, and extracted with ether ( $3 \times 30 \mathrm{ml}$ ). The combined organic layer was dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, $10 / 1$ ) to give $\mathbf{1 3}(751 \mathrm{mg}, 84 \%)$ as a colorless oil; $[\alpha]_{D}^{25}$ +43.0 (c 0.56, ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr:} \delta 0.88(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}), 0.92$ (t, 3H, J = 7.3 Hz ), 1.10-1.41 (m, 2H), 1.70-1.93 (m, 4H), 2.05$2.15(\mathrm{~m}, 1 \mathrm{H}), 2.70-2.92(\mathrm{~m}, 4 \mathrm{H}), 3.62-3.68(\mathrm{~m}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H})$, $4.13(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=4.2 \mathrm{~Hz}, \mathrm{~J}=10.2 \mathrm{~Hz}), 4.39(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0$ $\mathrm{Hz}), 4.54(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 6.88(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}), 7.29(\mathrm{~d}$, $2 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr:} \delta 12.1(\mathrm{q}), 13.6(\mathrm{q}), 25.8(\mathrm{t}), 26.1(\mathrm{t})$, 29.9 (t), 30.5 (t), 35.8 (t), 36.8 (d), 44.5 (d), 55.3 (q), 71.3 (t), 78.5 (d), 113.7 (d), 129.4 (d), 131.2 (s), 159.1 (s). MS (EI) m/z 340 $\left(\mathrm{M}^{+}, 5 \%\right), 219$ (74), 133 (100). hrms: Calcd. for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{2} \mathrm{~S}_{2}$ : $(\mathrm{M})^{+} 340.1531$. Found: m/z 340.1509.

## (3R,4S)-3-(4-Methoxybenzyloxy)-4-methylhexanal (14).

To a solution of $\mathbf{1 3}$ ( $1.10 \mathrm{~g}, 3.24$ mmoles) in aqueous $80 \%$ acetonitrile ( 30 ml ) was added mercury (II) chloride $(2.19 \mathrm{~g}, 8.10$ mmoles) and mercury (II) oxide ( $1.75 \mathrm{~g}, 8.10$ mmoles) at room temperature. The mixture was heated at $60^{\circ}$ with stirring for 2 hours. The cooled mixture was filtered through celite, and the filtrate was dried and concentrated. The residue was purified by chromatography (hexane/ethyl acetate, 10:1) to give 14 ( $680 \mathrm{mg}, 84 \%$ ) as a colorless oil; $[\alpha]_{\mathrm{D}}^{23}+18.5$ (c 0.73 , ethanol); ir (film) $1724 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ $\mathrm{nmr}: \delta 0.91(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}), 0.94(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.08-1.23$
$(\mathrm{m}, 1 \mathrm{H}), 1.34-1.48(\mathrm{~m}, 1 \mathrm{H}), 1.77-1.88(\mathrm{~m}, 1 \mathrm{H}), 2.42(\mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}=$ $1.7 \mathrm{~Hz}, \mathrm{~J}=3.5 \mathrm{~Hz}, \mathrm{~J}=16.3 \mathrm{~Hz}), 2.60(\mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}=2.7 \mathrm{~Hz}, \mathrm{~J}=$ $8.8 \mathrm{~Hz}, \mathrm{~J}=16.3 \mathrm{~Hz}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.85-3.90(\mathrm{~m}, 1 \mathrm{H}), 4.42(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}$ $=11.0 \mathrm{~Hz}), 4.52(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 6.88(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}), 7.24$ $(\mathrm{d}, 2 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}), 9.78(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=1.8 \mathrm{~Hz}, \mathrm{~J}=8.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{nmr}$ : $\delta 11.8(\mathrm{q}), 13.6(\mathrm{q}), 25.6(\mathrm{t}), 36.9(\mathrm{~d}), 44.3(\mathrm{t}), 55.2(\mathrm{q}), 70.9(\mathrm{t}), 77.3$ (d), 113.7 (d), 129.3 (d), 130.3 (s), 159.1 (s), 202.2 (d). hrms: Calcd. for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{3}:(\mathrm{M})+250.1569$. Found: $\mathrm{m} / \mathrm{z} 250.1545$.
$(3 S, 4 S)$ and $(3 R, 4 S)-3-\left[3^{\prime}, 3 '\right.$-(Ethylenedioxy)-3'-phenylpropanoyl]-3-methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone (17a) and (17b).

To a solution of lithium diisopropylamide, prepared from diisopropylamine $(0.12 \mathrm{ml}, 0.86 \mathrm{mmol})$ in tetrahydrofuran $(10 \mathrm{ml})$ and n-butyllithium ( 1.6 M in hexane, $0.49 \mathrm{ml}, 0.79 \mathrm{mmol}$ ), was added a solution of $\mathbf{1 5}(123 \mathrm{mg}, 0.66 \mathrm{mmol})$ in tetrahydrofuran $(5 \mathrm{ml})$ at $78^{\circ}$ under nitrogen, and the mixture was stirred for 30 minutes. A solution of $\mathbf{8}(165 \mathrm{mg}, 0.86 \mathrm{mmol})$ in tetrahydrofuran $(5 \mathrm{ml})$ was added to the reaction mixture. After stirring for 3.5 hours at $-78^{\circ}$, the mixture was quenched by addition of saturated aqueous ammonium chloride, and extracted with ether ( $4 \times 10 \mathrm{ml}$ ). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 20:1) to give 16 ( $207 \mathrm{mg}, 83 \%$ ) as an inseparable mixture of four diastereomers; ir (film) $3500,1765,1620 \mathrm{~cm}^{-1}$; To a solution of $16(37 \mathrm{mg})$ in acetone $(30 \mathrm{ml})$ was added Jones reagent $(8 \mathrm{M}, 0.04 \mathrm{ml})$ at $0^{\circ}$. After stirring for 1.5 hours, the mixture was quenched by addition of 2-propanol, followed by solid sodium bicarbonate. The mixture was filtered through celite and the filtrate was concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, $2: 1)$ to give $\mathbf{1 7 a}(14 \mathrm{mg}, 38 \%)$ and $\mathbf{1 7 b}(20 \mathrm{mg}, 54 \%)$ as colorless needles, respectively. 17a: mp $76^{\circ} ;[\alpha]_{\mathrm{D}}^{23}-29.3$ (c 0.40, ethanol); ir (nujol) 1770, $1715 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.28(\mathrm{~s}, 3 \mathrm{H}), 3.17(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $17.2 \mathrm{~Hz}), 3.39(\mathrm{~s}, 3 \mathrm{H}), 3.48(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.2 \mathrm{~Hz}), 3.73-3.85(\mathrm{~m}$, $2 \mathrm{H}), 4.00(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}), 4.02-4.12(\mathrm{~m}, 2 \mathrm{H}), 4.19(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $8.8 \mathrm{~Hz}), 4.97(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}), 5.05(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}), 5.12(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=17.0 \mathrm{~Hz}), 5.17(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.58(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=$ $10.8 \mathrm{~Hz}, \mathrm{~J}=17.4 \mathrm{~Hz}), 7.27-7.37(\mathrm{~m}, 3 \mathrm{H}), 7.48-7.52(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 18.1$ (q), 48.7 (t), 50.6 (s), 56.7 (q), $64.8(\mathrm{t} \mathrm{x} \mathrm{2)}$,73.6 ( t$)$, 88.8 (s), 94.6 (t), 108.3 (s), 117.7 (t), 125.8 (d), 128.1 (d), 128.2 (d), 135.7 (d), 141.9 (s), 171.6 (s), 203.0 (s). MS m/z $377\left[(\mathrm{M}+\mathrm{H})^{+}\right.$, 5\%], 317 (11), 271 (59), 185 (91), 149 (100), 93 (78). hrms: Calcd. for $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{O}_{7}:(\mathrm{M}+\mathrm{H})^{+} 377.1600$. Found: m/z 377.1579.

Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{7}$ : C, 63.82; H, 6.43. Found: C, 63.75; H, 6.47.

Compound 17b has mp $76^{\circ} ;[\alpha]_{\mathrm{D}}^{23}-30.2$ (c 2.0, ethanol); ir (nujol) 1770, $1715 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ : $\delta 1.00(\mathrm{~s}, 3 \mathrm{H}), 3.17(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $17.3 \mathrm{~Hz}), 3.35(\mathrm{~s}, 3 \mathrm{H}), 3.37(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.3 \mathrm{~Hz}), 3.74-3.82(\mathrm{~m}$, $2 \mathrm{H}), 3.83(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.5 \mathrm{~Hz}), 3.99-4.13(\mathrm{~m}, 2 \mathrm{H}), 4.29(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $8.5 \mathrm{~Hz}), 4.79(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz}), 4.85(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz}), 5.17$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=17.6 \mathrm{~Hz}), 5.33(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 6.15(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=$ $11.0 \mathrm{~Hz}, \mathrm{~J}=17.6 \mathrm{~Hz}), 7.26-7.35(\mathrm{~m}, 3 \mathrm{H}), 7.47-7.52(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 17.7$ (q), 48.7 (t), 49.3 ( s), 56.5 (q), 64.7 (t), 64.9 (t), 74.6 $(\mathrm{t}), 87.3(\mathrm{~s}), 94.6(\mathrm{t}), 108.1(\mathrm{~s}), 117.7(\mathrm{t}), 125.8(\mathrm{~d}), 128.0(\mathrm{~d})$, 128.1 (d), 135.2 (d), 141.9 (s), 170.2 (s), 200.7 (s). hrms: Calcd. for $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{O}_{7}:(\mathrm{M}+\mathrm{H})^{+} 377.1600$. Found: m/z 377.1583.
(-)-Hyperolactone C (3).
To a solution of $\mathbf{1 7 a}(9 \mathrm{mg}, 0.024 \mathrm{mmol})$ in tetrahydrofuran (10 ml ) was added a few drops of 3 M hydrochloric acid, and the mixture was heated at reflux for 11 hours. The mixture was
cooled and poured into aqueous sodium bicarbonate, and extracted with ether ( 3 x 20 ml ). The combined extracts were dried and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 10:1) to give a crystalline 3 (6 mg, 93\%); mp 106 ${ }^{\circ}$; (lit [2] mp 104 ${ }^{\circ}$ ); $[\alpha]_{D}^{25}-390$ (c 0.018, ethanol) [lit [2] -356 (c 0.018, ethanol)]; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.53(\mathrm{~s}, 3 \mathrm{H})$, $4.11(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 4.97(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 5.26(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=17.6 \mathrm{~Hz}), 5.27(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.98(\mathrm{~s}, 1 \mathrm{H}), 5.99(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.6 \mathrm{~Hz}), 7.49-7.54(\mathrm{~m}, 2 \mathrm{H}), 7.58-7.63(\mathrm{~m}$, $1 \mathrm{H}), 7.83-7.86(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 19.6$ (q), $48.9(\mathrm{~s}), 74.1(\mathrm{t})$, 93.1 ( s ), 100.3 (d), 119.0 ( t), 127.4 (d), 127.7 (s), 129.0 (d), 133.6 (d), 134.3 (d), 168.0 (s), 187.2 (s), 196.6 (s). MS (EI) m/z 270 ( $\mathrm{M}^{+}, 35 \%$ ), 252 (8), 225 (15), 211 (23), 197 (4), 187 (100), 173 (11), 147 (15), 105 (5), 129 (9), 102 (56). hrms: Calcd. for $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{4}:(\mathrm{M})^{+} 270.0892$. Found: m/z 270.0909.

## 5-Epi-hyperolactone C (3a).

The similar reaction of the $\beta$-methoxymethyloxy isomer $\mathbf{1 7 b}$ $(41 \mathrm{mg})$ as in the synthesis of $\mathbf{3}$ gave, after chromatography, 3a $(29.4 \mathrm{mg}, 100 \%)$ as a wax; $[\alpha]_{\mathrm{D}}^{20}+283$ (c 0.054 , ethanol); ${ }^{1} \mathrm{H}$ nmr: $\delta 1.30(\mathrm{~s}, 3 \mathrm{H}), 4.38(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.78(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6$ $\mathrm{Hz}), 5.34(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4 \mathrm{~Hz}), 5.37(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 6.04(\mathrm{~s}$, $1 \mathrm{H}), 6.06(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.4 \mathrm{~Hz}), 7.48-7.54(\mathrm{~m}, 2 \mathrm{H})$, 7.58-7.64 (m, 1H), 7.83-7.86 (m, 2H); ${ }^{13} \mathrm{C}$ nmr: $\delta 15.4$ (q), 48.9 ( s$), 73.3(\mathrm{t}), 92.3(\mathrm{~s}), 100.4(\mathrm{~d}), 116.2(\mathrm{t}), 127.4(\mathrm{~s}), 127.5(\mathrm{~d})$, 129.0 (d), 133.6 (d), 136.8 (d), 168.1 (s), 187.4 (s), 196.6 (s). MS (EI) m/z 270 ( $\mathrm{M}^{+}, 46 \%$ ), 252 (8), 211 (25), 187 (100), 173 (16), 147 (14), 129 (9), 102 (58). hrms: Calcd. for $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{4}$ : (M) ${ }^{+}$ 270.0892. Found: m/z 270.0880.
(3S, $4 S, 3 ' R, 4 ' S$ ) and (3R,4S, 3'R,4'S)-3-[3'-(4-Methoxy-benzyloxy)-4'-methylhexanoyl]-3-methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone (18a) and (18b).

The procedure for 16 was employed with 14 and 15 gave an alcohol as an inseparable mixture of four diastereomers in $84 \%$ yield. To a solution of the resulting alcohol ( $269 \mathrm{mg}, 0.62 \mathrm{mmol}$ ) in dichloromethane ( 20 ml ) was added Dess-Martin periodinane (915 $\mathrm{mg}, 2.16$ mmoles), and the mixture was stirred for 2 hours at room temperature. The reaction was quenched by addition of 2-propanol, and the precipitate was filtered off. The filtrate was washed with $10 \%$ sodium thiosulfate solution, and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 20:1) to give $\mathbf{1 8 a}(134 \mathrm{mg}, 50 \%)$ and $\mathbf{1 8 b}$ ( $134 \mathrm{mg}, 50 \%$ ). 18a (colorless oil); $[\alpha]_{\mathrm{D}}^{21}-0.20$ (c 2.7, ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.89(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=$ $6.8 \mathrm{~Hz}), 0.91(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.03-1.18(\mathrm{~m}, 1 \mathrm{H}), 1.30(\mathrm{~s}, 3 \mathrm{H})$, $1.23-1.40(\mathrm{~m}, 1 \mathrm{H}), 1.68-1.80(\mathrm{~m}, 1 \mathrm{H}), 2.59(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.7 \mathrm{~Hz}, \mathrm{~J}$ $=19.0 \mathrm{~Hz}), 2.99(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}, \mathrm{~J}=19.0 \mathrm{~Hz}), 3.43(\mathrm{~s}, 3 \mathrm{H})$, $3.79(\mathrm{~s}, 3 \mathrm{H}), 3.94-3.97(\mathrm{~m}, 1 \mathrm{H}), 4.00(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.18(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.41(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 4.48(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8$ $\mathrm{Hz}), 4.99(\mathrm{~s}, 2 \mathrm{H}), 5.11(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.14(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4$ $\mathrm{Hz}), 5.75(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.4 \mathrm{~Hz}), 6.84(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.4$ $\mathrm{Hz}), 7.22(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 12.0(\mathrm{q}), 13.9(\mathrm{q}), 17.7$ (q), 25.6 (t), 37.1 (d), 41.6 (t), 50.7 ( s), $55.2(\mathrm{q}), 56.6(\mathrm{q}), 71.6(\mathrm{t})$, 73.9 (t), 77.3 (d), 88.8 ( s$), 94.5(\mathrm{t}), 113.5(\mathrm{~d}), 117.5(\mathrm{t}), 129.2(\mathrm{~d})$, 131.1 ( s), 135.7 (d), 158.9 (s), 171.8 (s), 206.4 (s). hrms: Calcd. for $\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{7}:(\mathrm{M})^{+} 434.2304$. Found: m/z 434.2328. 18b (colorless oil); $[\alpha]_{\mathrm{D}}^{22}-11.3$ (c 3.76, ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.88$ (d, $3 \mathrm{H}, \mathrm{J}=6.8$ $\mathrm{Hz}), 0.91(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.11(\mathrm{~s}, 3 \mathrm{H}), 1.23-1.41(\mathrm{~m}, 2 \mathrm{H}), 1.70-$ $1.78(\mathrm{~m}, 1 \mathrm{H}), 2.31(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.8 \mathrm{~Hz}, \mathrm{~J}=18.1 \mathrm{~Hz}), 3.00(\mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}=9.4 \mathrm{~Hz}, \mathrm{~J}=18.1 \mathrm{~Hz}), 3.35(\mathrm{~s}, 3 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $8.4 \mathrm{~Hz}), 3.96-4.03(\mathrm{~m}, 1 \mathrm{H}), 4.36(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 4.41(\mathrm{~d}, 1 \mathrm{H}$,
$\mathrm{J}=10.6 \mathrm{~Hz}), 4.45(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.6 \mathrm{~Hz}), 4.77(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz})$, $4.81(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz}), 5.17(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.8 \mathrm{~Hz}), 5.34(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $11.0 \mathrm{~Hz}), 6.23(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}, \mathrm{~J}=17.8 \mathrm{~Hz}), 6.84(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=$ $8.6 \mathrm{~Hz}), 7.22(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 12.0(\mathrm{q}), 13.9(\mathrm{q}), 18.3$ (q), $26.0(\mathrm{t}), 37.0(\mathrm{~d}), 41.4(\mathrm{t}), 49.5(\mathrm{~s}), 55.2(\mathrm{q}), 56.5(\mathrm{q}), 72.0(\mathrm{t})$, 74.9 (t), 77.9 (d), 87.2 (s), 94.7 (t), 113.6 (d), 117.6 (t), 129.5 (d), 130.8 (s), 135.1 (d), 159.0 (s), 170.7 (s), 204.8 (s). hrms: Calcd. for $\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{7}:(\mathrm{M})^{+} 434.2304$. Found: $\mathrm{m} / \mathrm{z} 434.2281$.
( $3 S, 4 S, 3^{\prime} R, 4^{\prime} S$ ) and ( $3 R, 4 S, 3^{\prime} R, 4^{\prime} S$ )-3-(3'-Hydroxy-4'-methyl-hexanoyl)-3-methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone (19a) and (19b).

A mixture of $\mathbf{1 8 a}(131 \mathrm{mg}, 0.3 \mathrm{mmol})$, dichlorodicyanobenzoquinone (DDQ) ( $177 \mathrm{mg}, 0.76 \mathrm{mmol}$ ) in dichloromethane $(9 \mathrm{ml})$ and water $(0.5 \mathrm{ml})$ was stirred at room temperature for 90 minutes. The mixture was diluted with saturated aqueous sodium bicarbonate solution and extracted with dichloromethane ( $3 \times 30 \mathrm{ml}$ ). The organic layer was dried and concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, 10:1) to give 19a ( $84 \mathrm{mg}, 94 \%$ yield) as a colorless oil; $[\alpha]_{\mathrm{D}}^{23}+1.87$ (c 1.65, ethanol); ir (film) 3545, 1790, $1710 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.87(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.6$ $\mathrm{Hz}), 0.90(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 1.04-1,26(\mathrm{~m}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.43-$ $1.57(\mathrm{~m}, 2 \mathrm{H}), 2.74(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}), 2.79(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=4.2 \mathrm{~Hz}$, $\mathrm{J}=15.0 \mathrm{~Hz}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.88-3.95(\mathrm{~m}, 1 \mathrm{H}), 4.12(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.8$ $\mathrm{Hz}), 4.25(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.8 \mathrm{~Hz}), 4.96(\mathrm{~s}, 2 \mathrm{H}), 5.23(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4$ $\mathrm{Hz}), 5.29(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.78(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.4$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 11.4$ (q), 14.4 (q), 17.0 (q), 24.8 (t), 39.7 (d), 44.3 $(\mathrm{t}), 50.6(\mathrm{~s}), 56.6(\mathrm{q}), 70.8(\mathrm{~d}), 74.0(\mathrm{t}), 87.7(\mathrm{~s}), 94.5(\mathrm{t}), 117.7(\mathrm{t})$, 135.9 (d), $171.5(\mathrm{~s}), 208.8(\mathrm{~s})$. hrms (FAB): Calcd. for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{6} \mathrm{Na}^{+}:(\mathrm{M}+\mathrm{Na})^{+} 337.1627$. Found: m/z 337.1640.

The $\beta$-methoxymethyloxy isomer $\mathbf{1 8 b}$ was treated as in the synthesis of 19a to give a colorless oil 19b (94\% yield); $[\alpha]_{D}^{23}$ -11.1 (c 2.55, ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr:} \delta 0.87$ (d, 3H, J $\left.=6.6 \mathrm{~Hz}\right), 0.90$ $(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.14(\mathrm{~s}, 3 \mathrm{H}), 1.10-1.17(\mathrm{~m}, 1 \mathrm{H}), 1.41-1.56$ $(\mathrm{m}, 2 \mathrm{H}), 2.63-2.77(\mathrm{~m}, 2 \mathrm{H}), 2.92(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=3.3 \mathrm{~Hz}, \mathrm{OH}), 3.42$ $(\mathrm{s}, 3 \mathrm{H}), 3.90(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 3.90-3.96(\mathrm{~m}, 1 \mathrm{H}), 4.36(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.86(\mathrm{~s}, 2 \mathrm{H}), 5.24(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.6 \mathrm{~Hz}), 5.38$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 6.25(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.6 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ nmr: $\delta 11.4$ (q), 14.4 (q), 18.2 (q), 24.8 (t), 39.5 (d), 43.6 (t), 49.3 ( s$), 56.6(\mathrm{q}), 70.6(\mathrm{~d}), 74.8(\mathrm{t}), 87.1(\mathrm{~s}), 94.6(\mathrm{t}), 118.0$ (t), $134.9(\mathrm{~d}), 170.2(\mathrm{~s}), 207.3(\mathrm{~s})$. hrms (FAB): Calcd. for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{6} \mathrm{Na}^{+}:(\mathrm{M}+\mathrm{Na})^{+} 337.1627$. Found: m/z 337.1640.
( $3 S, 4 \mathrm{~S}, 4$ 'S) and (3R, $4 S, 4$ 'S)-3-(3'-Oxo-4'-methylhexanoyl)-3-methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone (20a) and (20b).

To a solution of $19 \mathbf{a}(28.1 \mathrm{mg}, 0.09 \mathrm{mmol})$ in acetone ( 5 ml ) was added an excess amount of Jones reagent ( $8 M$ solution) at $-20^{\circ}$. After stirring for 2 hours, the reaction mixture was quenched by addition of 2-propanol and solid sodium bicarbonate. The mixture was filtered through Celite, and the filtrate was concentrated in vacuo. The residue was purified by chromatography (hexane/ethyl acetate, 9:1) to give 20a ( $18.3 \mathrm{mg}, 66 \%$ yield) as a pale yellow oil; $[\alpha]_{\mathrm{D}}^{21}-11.7$ (c 0.63, ethanol); ir (film) 2970, $1790,1600 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.89(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}), 1.14(\mathrm{~d}, 3 \mathrm{H}$, $\mathrm{J}=7.0 \mathrm{~Hz}), 1.36(\mathrm{~s}, 3 \mathrm{H}), 1.42-1.54(\mathrm{~m}, 1 \mathrm{H}), 1.59-1.73(\mathrm{~m}, 1 \mathrm{H})$, 2.22-2.34 (m, 1H), $3.47(\mathrm{~s}, 3 \mathrm{H}), 4.12(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.33(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 4.94(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=5.9 \mathrm{~Hz}), 5.09(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=5.9$ $\mathrm{Hz}), 5.13(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4 \mathrm{~Hz}), 5.16(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 5.73$ $(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}, \mathrm{~J}=17.4 \mathrm{~Hz}), 5.92(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 11.6$
(q), $17.2(\mathrm{q}), 17.9(\mathrm{q}), 27.1(\mathrm{t}), 43.6(\mathrm{~d}), 50.7(\mathrm{~s}), 56.8(\mathrm{q}), 73.9(\mathrm{t})$, 85.8 ( s ), 94.4 ( t$), 97.8$ (d), 116.3 ( t), 136.4 (d), 172.2 ( s$), 190.8$ ( s$)$, 196.5 (s). hrms: Calcd. for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{6}:(\mathrm{M})^{+} 312.1573$. Found: $\mathrm{m} / \mathrm{z}$ 312.1563.

The $\beta$-methoxymethyloxy isomer $\mathbf{1 9 b}$ was treated as in the synthesis of 20a to give pale yellow crystals $\mathbf{2 0 b}$ ( $65 \%$ ); mp 54.5 ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{20}+13.6$ (c 0.89, ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.90(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.5$ $\mathrm{Hz}), 1.15(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}), 1.16(\mathrm{~s}, 3 \mathrm{H}), 1.41-1.55(\mathrm{~m}, 1 \mathrm{H})$, $1.59-1.73(\mathrm{~m}, 1 \mathrm{H}), 2.25-2.37(\mathrm{~m}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=8.4 \mathrm{~Hz}), 4.43(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 4.83(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz})$, $4.92(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}), 5.17(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.6 \mathrm{~Hz}), 5.32(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=11.0 \mathrm{~Hz}), 5.95(\mathrm{~s}, 1 \mathrm{H}), 6.17(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}, \mathrm{~J}=17.6$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ nmr: $\delta 11.6$ (q), 17.0 (q), 18.9 (q), 27.1 (t), 43.8 (d), 49.9 ( s$), 56.7(\mathrm{q}), 74.8(\mathrm{t}), 84.9(\mathrm{~s}), 94.4(\mathrm{t}), 98.2(\mathrm{~d}), 116.9(\mathrm{t}), 135.6$ (d), 171.2 (s), 188.2 (s), 197.9 (s).

## (2'S,5S,9S)-Isomer (1): (-)-Hyperolactone A.

To a cooled solution of $\mathbf{2 0 a}(7.5 \mathrm{mg})$ in dichloromethane ( 5 ml ) was added trimethylsilyl bromide $(0.03 \mathrm{ml}, 0.22 \mathrm{mmol})$ at $-20^{\circ}$, and the mixture was stirred for 14 hours. The reaction was quenched by addition of saturated aqueous sodium bicarbonate and the mixture was extracted with dichloromethane ( 3 x 10 ml ). The combined extracts were washed with water, dried and concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, $10: 1)$ to give $\mathbf{1}(5.9 \mathrm{mg}, 98 \%)$ as a single product; mp $57^{\circ}$ (recrystallized from ether and $n$-hexane); [lit [1] $\mathrm{mp} 57^{\circ}$ for natural hyperolactone A]; $[\alpha]_{D}^{24}-265$ (c 0.13, methanol); [lit [1] $[\alpha]_{D}^{24}-229$ (c 0.13 , methanol)]; ${ }^{1} \mathrm{H}$ nmr: $\delta 0.96(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}), 1.25(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}$ $=6.8 \mathrm{~Hz}), 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.59-1.78(\mathrm{~m}, 2 \mathrm{H}), 2.63-2.75(\mathrm{~m}, 1 \mathrm{H}), 4.05$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 4.90(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 5.24(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4$ $\mathrm{Hz}), 5.28(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.38(\mathrm{~s}, 1 \mathrm{H}), 5.94(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8$ $\mathrm{Hz}, \mathrm{J}=17.4 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{nmr}$ : $\delta 11.0(\mathrm{q}), 16.9(\mathrm{q}), 19.3(\mathrm{q}), 27.1(\mathrm{t})$, 37.1 (d), 48.4 ( s$), 74.0(\mathrm{t}), 92.5$ ( s$), 102.1$ (d), 118.8 (t), 134.3 (d), 168.0 (s), 197.3 (s), 200.2 (s). MS (EI) m/z 250 (M+, 31\%), 193 (53), 177 (42), 167 (100). hrms: Calcd. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}:(\mathrm{M})^{+}$ 250.1205. Found: m/z 250.1216.

## ( 2 'S,5R,9S)-Isomer (1a).

The $\beta$-methoxymethyloxy isomer 20b was treated as in the synthesis of $\mathbf{1}$ to give a colorless wax $\mathbf{1 a}$ ( $89 \%$ yield) as a single product; $[\alpha]_{\mathrm{D}}^{21}+194$ (c 0.20 , ethanol); ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.99(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $7.3 \mathrm{~Hz}), 1.23(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}), 1.54-1.68(\mathrm{~m}, 1 \mathrm{H})$, $1.67-1.79(\mathrm{~m}, 1 \mathrm{H}), 2.61-2.73(\mathrm{~m}, 1 \mathrm{H}), 4.31(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz})$, $4.71(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}), 5.27(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.4 \mathrm{~Hz}), 5.30(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=10.8 \mathrm{~Hz}), 5.43(\mathrm{~s}, 1 \mathrm{H}), 5.93(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.4 \mathrm{~Hz}) ;$ ${ }^{13} \mathrm{C} \mathrm{nmr:} \delta 11.3$ (q), 15.4 (q), 17.2 (q), 27.1 (t), 37.3 (d), 48.4 (s), 73.2 (t), 90.3 ( s$), 102.2$ (d), 116.0 ( t), 136.8 (d), 168.0 (s), 197.4 (s), 200.5 (s). MS (EI) m/z $250\left(\mathrm{M}^{+}, 23 \%\right), 193$ (42), 177 (47), 167 (100). hrms: Calcd. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}:(\mathrm{M})^{+} 250.1205$. Found: m/z 250.1215 .

## ( $2^{\prime} R, 5 S, 9 S$ )-Isomer (1b).

The $\alpha$-methoxymethyloxy isomer 20a was treated with boiling tetrahydrofuran containing 3 M hydrochloric acid to give a mixture of $\mathbf{1}$ and its $2^{\prime}$-isomer $\mathbf{1 b}$ ( $88 \%$ yield), which was separated by HPLC [Shimadzu LC-3A; column: Cosmosil 5SI; n-hexanediethyl ether (95:5); $5 \mathrm{ml} /$ minute]. The first peak (Rt $21.25 \mathrm{~min}-$ utes) gave 1 and second peak (Rt 21.75 minutes) gave $\mathbf{1 b}$ as a colorless wax. 1b: ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 0.97(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.25(\mathrm{~d}$, $3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.50-1.62(\mathrm{~m}, 1 \mathrm{H}), 1.62-1.80(\mathrm{~m}$, $1 \mathrm{H}), 2.69($ sextet, $1 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}), 4.04(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.5 \mathrm{~Hz}), 4.89$
(d, 1H, J = 8.5 Hz ), $5.24(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.1 \mathrm{~Hz}), 5.28(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ 11.0 Hz ), 5.37 (s, 1H), 5.93 (dd, 1H, J = 11.0 Hz, J = 17.1 Hz ); ${ }^{13} \mathrm{C}$ nmr: $\delta 11.3$ (q), 17.3 (q), 19.3 (q), 27.2 (t), 37.3 (d), 48.4 (s), 74.0 (t), 92.5 ( s ), 102.2 (d), 118.9 (t), 134.4 (d), 168.0 ( s$), 197.4$ (s), 200.2 (s). hrms: Calcd. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}:(\mathrm{M})^{+} 250.1205$. Found: m/z 250.1195.

Preparation of the Key Intermediate 15.
(2S,3R)-5-Benzyloxy-3-ethoxycarbonyl-2-hydroxy-3-methylpentanoic acid (21b).

To a solution of lithium diisopropylamide, prepared from diisopropylamine ( $11.3 \mathrm{ml}, 80.4$ mmoles) and $n$-butyllithium ( 1.6 M in hexane, $45.6 \mathrm{ml}, 73.7 \mathrm{mmoles}$ ) in tetrahydrofuran ( 200 ml ), was added a solution of diethyl ( $2 S, 3 R$ )-(+)-2-hydroxy-3-methylsuccinate [8] ( $6.8 \mathrm{~g}, 33.4$ mmoles) in tetrahydrofuran ( 12 ml ) at $-78^{\circ}$ and the mixture was stirred for 3 hours. To the cooled ( -78 ) solution was added benzyl 2-iodoethyl ether [9] ( $13.1 \mathrm{~g}, 50$ mmoles) within 5 minutes and the mixture was stirred for 8 hours, then at $0^{\circ}$ for 13 hours. After quenching by addition of a solution of acetic acid (8 ml ) in 15 ml of ether, the mixture was poured into water and extracted with ether ( $4 \times 100 \mathrm{ml}$ ). The combined extracts were washed with aqueous sodium bicarbonate, dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 10:1) to give 21a ( $12.46 \mathrm{~g}, 67 \%$ ) as a pale yellow oil; $[\alpha]_{\mathrm{D}}^{27}+12.9$ (c 0.70, ethanol); ir (film) $3490,1715 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ : $\delta 1.21(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 1.22(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $7.1 \mathrm{~Hz}), 1.80-1.99(\mathrm{~m}, 1 \mathrm{H}), 2.10-2.22(\mathrm{~m}, 1 \mathrm{H}), 3.55(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.5$ $\mathrm{Hz}), 3.65(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}, \mathrm{OH}), 4.12(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 4.24$ $(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 4.32(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 4.47(\mathrm{~s}, 2 \mathrm{H}), 7.27-$ 7.33 (m, 5H); MS (EI) m/z 338 (M+ $15 \%$ ), 293 (4), 231 (16), 204 (23), 157 (35), 131 (30), 91 (100); hrms: Calcd. for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{O}_{6}$ : $(\mathrm{M})^{+} 338.1729$. Found: $\mathrm{m} / \mathrm{z} 338.1728$. To the resulting diester 21a $(5.11 \mathrm{~g})$ in methanol ( 40 ml ) was added a solution of potassium hydroxide ( $2.85 \mathrm{~g}, 3.4$ equivalents) in methanol ( 30 ml ) and water $(2 \mathrm{ml})$ at $-40^{\circ}$, and the mixture was stirred for 1 hour, then at room temperature for 3 days. The mixture was acidified by addition of 3 $M$ hydrochloric acid to $\mathrm{pH}-2$ and extracted with ether ( $4 \times 100 \mathrm{ml}$ ). The combined extracts were dried and concentrated in vacuo to give 21b ( $4.55 \mathrm{~g}, 97 \%$ ) as a pale yellow syrup; $[\alpha]_{\mathrm{D}}^{27}+8.4$ (c 0.71 , ethanol); ir (film) $3400,1700 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.21$ (t, 3H, J = 7.1 $\mathrm{Hz}), 1.25(\mathrm{~s}, 3 \mathrm{H}), 1.99-2.15(\mathrm{~m}, 2 \mathrm{H}), 3.57-3.66(\mathrm{~m}, 2 \mathrm{H}), 4.13(\mathrm{q}$, $2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 4.51(\mathrm{~s}, 2 \mathrm{H}), 7.27-7.33(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 13.9$ (q), $18.6(\mathrm{q}), 34.8(\mathrm{t}), 49.0(\mathrm{~s}), 60.9(\mathrm{t}), 61.4(\mathrm{t}), 66.4(\mathrm{t}), 73.4(\mathrm{t})$, 74.4 (d), 128.0 (d), 128.5 (d x 2), 137.0 (s), 174.4 (s), 174.9 (s). hrms: Calcd. for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{6}:(\mathrm{M})^{+} 310.1416$. Found: $\mathrm{m} / \mathrm{z} 310.1429$.
(3S,4S)-4-(2-Benzyloxyethyl)-3-hydroxy-4-methyl- $\gamma$-butyrolactone (22a).

To a solution of $\mathbf{2 1 b}$ ( $2.29 \mathrm{~g}, 7.4$ mmoles) in tetrahydrofuran (100 ml ) was added lithium triethylborohydride ( 1.0 M in tetrahydrofuran, $44.3 \mathrm{ml}, 44.3 \mathrm{mmoles}$ ) at $-78^{\circ}$, and the mixture was stirred for 3 hours, then at room temperature for 40 hours. To the reaction mixture was added 17 ml of water at $-10^{\circ}$, and the aqueous layer was extracted with ether ( $3 \times 50 \mathrm{ml}$ ). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 2:1) to give 22a ( 1.37 g , $74 \%$ ) as a colorless oil; $[\alpha]_{\mathrm{D}}^{27}+6.0$ (c 0.80 , ethanol); ir (film) 3400 , $1760,1620 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr: $\delta 1.26(\mathrm{~s}, 3 \mathrm{H}), 1.58-1.94(\mathrm{~m}, 2 \mathrm{H}), 3.48$ (dt, 1H, J = 2.4 Hz, J = 10.1 Hz ), 3.57-3.66 (m, 1H), $3.97(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}$ $=9.3 \mathrm{~Hz}), 4.08(\mathrm{~s}, 1 \mathrm{H}), 4.19(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.3 \mathrm{~Hz}), 4.31(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $11.7 \mathrm{~Hz}), 4.59(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7 \mathrm{~Hz}), 7.30-7.38(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta$
23.5 (q), 33.8 ( t), 42.5 ( s$), 66.0$ ( t$), 73.1$ (t), 74.7 (t), 75.4 (d), 127.9 (d), 128.1 (d), 128.4 (d), 136.7 (s), 176.5 (s). hrms: Calcd. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}:(\mathrm{M})^{+} 250.1205$. Found: $\mathrm{m} / \mathrm{z} 250.1219$.
(3S,4S)-4-(2-Benzyloxyethyl)-3-methoxymethyloxy-4-methyl- $\gamma$ butyrolactone (22b).

To a solution of $\mathbf{2 2 a}$ ( $730 \mathrm{mg}, 2.92 \mathrm{mmoles}$ ) and diisopropylethylamine ( $5.1 \mathrm{ml}, 29.0$ mmoles) in dichloromethane ( 50 ml ) was added chloromethoxymethane $(1.11 \mathrm{ml}, 14.6 \mathrm{mmoles})$, and the mixture was heated at reflux for 48 hours. The mixture was cooled, diluted with water, and extracted with dichloromethane ( $4 \times 20 \mathrm{ml}$ ). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 3:1) to give 22b ( $795 \mathrm{mg}, 93 \%$ ) as a pale yellow oil; $[\alpha]_{\mathrm{D}}^{27}$ -47.7 (c 2.3, ethanol); ir (film) $1770 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ : $\delta 1.19$ (s, 3H), $1.72-1.95(\mathrm{~m}, 2 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}), 3.57-3.62(\mathrm{~m}, 2 \mathrm{H}), 3.88(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $9.0 \mathrm{~Hz}), 3.99(\mathrm{~s}, 1 \mathrm{H}), 4.39(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}), 4.47(\mathrm{~s}, 2 \mathrm{H}), 4.69(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}), 4.99(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}), 7.28-7.35(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ $\mathrm{nmr}: \delta 21.1$ (q), 31.3 (t), 42.1 ( s$), 56.0(\mathrm{q}), 66.4(\mathrm{t}), 73.2$ (t), 75.1 ( t$),$ 78.3 (d), 95.9 (t), 127.6 (d x 2), 128.4 (d), 138.1 (s), 175.0 (s). MS (EI): m/z 249 ([M-CH2 $\left.\mathrm{OCH}_{3}\right]^{+}, 40 \%$ ), 143 (65), 91 (100).
(3S,4S)-4-(2-Hydroxyethyl)-3-methoxymethyloxy-4-methyl- $\gamma$ butyrolactone (23a).

A mixture of $\mathbf{2 2 b}$ ( $622 \mathrm{mg}, 2.11 \mathrm{mmoles}$ ) and $10 \%$ palladium on carbon ( 300 mg ) in ethanol ( 25 ml ) was vigorously stirred under an atmosphere of hydrogen at room temperature for 44 hours. The mixture was filtered and the filtrate was concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 1:1) to give 23a ( $414 \mathrm{mg}, 96 \%$ ) as a colorless oil; $[\alpha]_{D}^{22}-94.6$ (c 0.65, ethanol); ir (film) $3425,1760 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.24(\mathrm{~s}, 3 \mathrm{H}), 1.79(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.6 \mathrm{~Hz}), 2.21(\mathrm{br} \mathrm{s}, 1 \mathrm{H}$, $\mathrm{OH}), 3.44(\mathrm{~s}, 3 \mathrm{H}), 3.70-3.82(\mathrm{~m}, 2 \mathrm{H}), 3.93(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz})$, $4.05(\mathrm{~s}, 1 \mathrm{H}), 4.35(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}), 4.73(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz})$, $5.01(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ : $\delta 21.1(\mathrm{q}), 34.6(\mathrm{t}), 42.1(\mathrm{~s})$, 56.0 (q), 58.7 (t), 75.0 (t), 78.3 (d), 96.0 (t), 175.0 (s). MS (EI) $\mathrm{m} / \mathrm{z}\left(\left[\mathrm{M}-\mathrm{OCH}_{2} \mathrm{OCH}_{3}\right]^{+}, 16 \%\right), 129$ (11), 113 (43), 99 (100), 91 (11), 85 (27). hrms: Calcd. for $\mathrm{C}_{7} \mathrm{H}_{11} \mathrm{O}_{3}:\left(\mathrm{M}-\mathrm{OCH}_{2} \mathrm{OCH}_{3}\right)^{+}$ 143.0708. Found: m/z 143.0722.
(3S,4S)-3-Methoxymethyloxy-4-methyl-4-[2-(2-nitrophenyl)-seleno]- $\gamma$-butyrolactone (23b).

A mixture of 23a ( $673 \mathrm{mg}, 3.3 \mathrm{mmoles}$ ), 2-nitrophenylselenocyanate ( $1.05 \mathrm{~g}, 4.62 \mathrm{mmoles}$ ) and $n$-tributylphosphine ( 1.15 ml , 4.62 mmoles) in 30 ml of tetrahydrofuran was stirred at room temperature for 1 hour. The reaction mixture was diluted with water and extracted with ether ( $5 \times 20 \mathrm{ml}$ ). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 4/1) to give 23b ( $1.13 \mathrm{~g}, 88 \%$ ) as yellow crystals; mp 74-75 ${ }^{\circ} ;[\alpha]_{\mathrm{D}}^{27}$ -37.0 (c 0.24, ethanol); ir (nujol) 1770, 1590, 1560, $1510 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}: \delta 1.29(\mathrm{~s}, 3 \mathrm{H}), 1.87-2.06(\mathrm{~m}, 2 \mathrm{H}), 2.85-3.00(\mathrm{~m}, 2 \mathrm{H})$, $3.46(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}), 4.18(\mathrm{~s}, 1 \mathrm{H}), 4.26(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=9.4 \mathrm{~Hz}), 4.78(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}), 5.08(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz})$, 7.30-7.35 (m, 1H), 7.50-7.59 (m, 3H), $8.28(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=1.3 \mathrm{~Hz}$, $\mathrm{J}=8.3 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{nmr}$ : $\delta 20.4(\mathrm{t}), 22.2(\mathrm{q}), 33.3(\mathrm{t}), 44.7(\mathrm{~s}), 56.2$ (q), 74.3 (t), 77.5 (d), 96.2 (t), 125.5 (d), 126.5 (d), 128.9 (d), 132.8 (d), 133.7 (s), 146.7 (s), 174.8 (s). hrms: Calcd. for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NO}_{6} \mathrm{Se}:(\mathrm{M})+389.0377$. Found: m/z 389.0399.

Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NO}_{6} \mathrm{Se}: \mathrm{C}, 46.40 ; \mathrm{H}, 4.93 ; \mathrm{N}, 3.61$. Found: C, 46.45; H, 4.98; N, 3.61.
(3S,4S)-3-Methoxymethyloxy-4-methyl-4-vinyl- $\gamma$-butyrolactone (15).

To a solution of $\mathbf{2 3 b}(1.11 \mathrm{~g}, 2.85$ mmoles) in tetrahydrofuran $(40 \mathrm{ml})$ was added dropwise $30 \%$ hydrogen peroxide ( 1.58 ml , 17.2 mmoles) at $0^{\circ}$, and the mixture was warmed to room temperature. After stirring for 5 hours, the reaction mixture was diluted with water and extracted with ether ( $4 \times 50 \mathrm{ml}$ ). The combined extracts were washed with sodium thiosulfate solution, dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 6:1) to give 15 (403 $\mathrm{mg}, 76 \%$ ) as an oil; $[\alpha]_{\mathrm{D}}^{25}-64.7$ (c 2.20, ethanol); ir (film) 1780 , $1630 \mathrm{~cm}^{-1}{ }^{1}{ }^{1} \mathrm{H} \mathrm{nmr}$ : $\delta 1.32(\mathrm{~s}, 3 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.98(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $9.0 \mathrm{~Hz}), 4.13(\mathrm{~s}, 1 \mathrm{H}), 4.28(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}), 4.71(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $6.8 \mathrm{~Hz}), 4.98(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}), 5.23(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=17.6 \mathrm{~Hz}), 5.27$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 5.97(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}, \mathrm{~J}=17.5 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ nmr: $\delta 20.4$ (q), 45.6 (s), 55.9 (q), 74.1 (t), 78.4 (d), 96.0 (t), 115.9 (t), 136.3 (d), 174.2 (s). hrms (CI): Calcd. for $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{O}_{4}$ : (M $+\mathrm{H})^{+}$187.0934. Found: m/z 187.0954.

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