Synthesis of Hyperolactones A and C

Toshihiko Ueki [a], Matsumi Doe [a], Rika Tanaka [b], Yoshiki Morimoto [a], Kazuo Yoshihara [c] and Takamasa Kinoshita* [a]

Department of Chemistry, Graduate School of Science, Osaka City University, Sumiyoshi-ku, Osaka 558-8585, Japan Department of Applied Chemistry, Faculty of Engineering, Osaka City University Suntory Institute for Bioorganic Research, Shimamoto-cho, Osaka 618-0024, Japan

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Hyperolactones A (1) and C (3) have been synthesized starting from (S)-malic acid by a straightforward route. The unique spirolactone skeleton was efficiently constructed by one-pot reaction as a key step. The absolute stereochemistry of hyperolactones was unambiguously established by this synthesis.

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(-)-Hyperolactone A (1) was isolated from the leaves and stems of the Chinese plant Hypericum chinense L. by M. Tada and co-workers in 1989 [1]. Later, two structural relatives of 1, (+)-hyperolactone B (2) and (-)-hyperolactone C (3), were also isolated from the same plant as minor components in 1995 [2]. The gross structure of 1 and its relative stereochemistry were determined by means of extensive spectroscopic studies and chemical transformation, and was finally confirmed by X-ray crystallographic analysis. These compounds possess a common spirolac-





Hyperolactone A (1)

Figure 1

tone skeleton (anti-orientation at the stereogenic center C-5 in 2 as compared with 1 and 3), and some novel characteristics of the structure are (1) a highly substituted 1,7dioxaspiro[4,4]non-2-ene-4,6-dione skeleton; (2) a strained 2,2,5-trisubstituted 3(2H)furanone ring system. Owing to their structural uniqueness, compound 1 and structurally related natural products are good targets for total synthesis [3]. We now describe herein the synthesis of 1 and 3 in a stereoselective fashion, which enables us to establish their absolute stereochemistry unambiguously.

Synthesis of the Side Chain Units 8 and 14.

The requisite aldehyde 8 was synthesized as shown in Scheme 1. The β -keto ester 6 was obtained by the Collins oxidation of the β -hydroxy ester 5, which was prepared from ethyl bromoacetate and benzaldehyde 4 by the Reformatsky reaction [4]. The keto ester 6 was heated with ethylene glycol containing a small amount of p-toluenesulfonic acid in benzene to give the ketal 7 in 73% yield. Diisobutylaluminium hydride reduction of 7 afforded the aldehyde 8 in 78% yield.



Reagents and conditions: i, Zn, BrCH₂CO₂CH₂CH₃, benzene, reflux, 85%; ii, CrO₃, pyridine, CH₂Cl₂, rt, 82%; iii, ethyleneglycol, p-TsOH, benzene, reflux, 73%; iv, diisobutylaluminium hydride, toluene, -78°, 78%



PMB=p-methoxybenzyl

Reagents and conditions: i, ref. 5; ii, LiAlH₄, tetrahydrofuran, reflux, 83%; iii, [PhCH₂NEt₃]⁺Cl⁻, MsCl, NaOH, CH₂Cl₂, 62%; iv, 1,3-dithiane, n-BuLi, tetrahydrofuran, -40° to 0°, 79%; v, NaH, p-methoxybenzyl chloride, dimethylformamide, 84%. vi, HgCl₂, HgO, CH₃CN, reflux, 84%.

Since the X-ray crystallographic analysis of **1** suggests (2'S)-configuration, a synthetic plan for the requisite (4S)aldehyde **14** as a coupling partner is shown in Scheme 2. Reduction of (2S, 3S)-acetoxy-3-methylvaleric acid **9** [5], prepared from (2S, 3S)-isoleucine, with lithium aluminium hydride afforded the diol **10** in 83% yield, which was converted [6] into the volatile epoxide **11** in 62% yield. The lithiation of 1,3-dithiane with *n*-butyllithium and subsequent addition to the epoxide **11** gave the dithiane **12** in 79% yield. *p*-Methoxybenzyl protection of the hydroxyl group in **12** and subsequent oxidative cleavage of the dithiane afforded the desired aldehyde **14**.

Synthesis of 3 (Scheme 3).

The aldol reaction of the lithium enolate of (3S,4S)-3methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone **15** (Scheme 5, see experimental section) [3b] with the aldehyde **8** afforded **16** as an inseparable mixture of four diastereomers in 83% yield. Although a multitude of isomers is formed, in the light of the subsequent steps this factor is of minor significance. Successive treatment of **16** with the Jones reagent provided **17** in 92% yield (a:b = 2:3) as a separable mixture.

The stereochemistries of 17 were confirmed by ¹H NMR analysis and NOE experiments of 17a and 17b

(Figure 2). Both in 17a and 17b, a NOE was observed between the 4-methyl group (4-CH₃) and a higher-field hydrogen (5-Ha) of the 5-methylene, but not observed between the 4-CH₃ and a lower-field hydrogen (5-Hb). In 17a, a small NOE was observed between the 4-CH₃ and a methylene hydrogen of the methoxymethyloxy group, whereas a NOE was not observed in 17b. By the anisotropic effect of the methoxymethyloxy group, the signals of 4-CH₃ and 5-Ha for 17a appeared in the lower field than the corresponding signals for 17b, and 5-Hb signal for 17b was observed in the lower field than that for 17a. These results indicate that the methoxymethyloxy group and the 4-CH₃ in **17a** were located in the syn relationship. From these considerations, it was found that the stereochemistry of 17a was (3S, 4S)- configuration and 17b was its 3R-epimer.

Acid-catalyzed hydrolysis of **17a** in boiling tetrahydrofuran containing 3 *M* hydrochloric acid furnished **3** (93% yield) consequently *via* a subsequent ring closure and dehydration (Scheme 3). The compound **3** was found to be identical with the natural product [2] by comparing the mp, $[\alpha]_D$, ms, ¹H and ¹³C NMR spectra with those reported, while the physical data of compound **3a** derived from **17b** were all different except the hrms.

Scheme 3



MOM = methoxymethyl

Reagents and conditions: i, lithium diisopropylamide, tetrahydrofuran, -78°, 83%; ii, CrO₃, H₂SO₄, acetone, 0°, 92%; iii, 3M HCl, tetrahydrofuran, reflux, **3** (93%); **3a** (100%).



Reagents and conditions: i, (a) lithium diisopropypamide, tetrahydrofuran, -78°, 84% (b) Dess-Martin periodinane, CH₂Cl₂, 100%; ii, dichlorodicyanobenzoquinone, CH₂Cl₂-H₂O, 94%; iii, CrO₃, H₂SO₄, acetone, -20°, 66%; iv,trimethylsilyl bromide, CH₂Cl₂, -20°, 98%.





ms, ¹H and ¹³C NMR spectra with those reported [1]. The β -methoxymethyloxy isomer **20b** was treated with bromotrimethylsilane to afford a *5R*-isomer **1a** as a single product in 95% yield.

Thus, we have completed the total synthesis of **1** and **3**. This synthesis discloses that the stereochemistry should be depicted as (2'*S*,5*S*,9*S*)-2-(2'-butyl)-9-ethenyl-9-methyl-1,7-dioxaspiro-[4,4]non-2-ene-4,6-dione for (-)-hyperolactone A and (5*S*,9*S*)-configuration for (-)-hyperolactone C, respectively.



Scheme 5

Reagents and conditions: i, lithium diisopropylamide, PhCH₂OCH₂CH₂I, tetrahydrofuran, -78°, 67%; ii, KOH, MeOH-H₂O, rt, 97%; iii, LiB(C₂H₅)₃H, tetrahydrofuran, rt, 74%; iv, CH₃OCH₂Cl,*i*-Pr₂EtN, CH₂Cl₂, reflux, 93%; v, 10% Pd-C, H₂, EtOH, 96%; vi, *o*-O₂NC₆H₄SeCN, Bu₃P, tetrahydrofuran, rt, 88%; vii, 30% H₂O₂, tetrahydrofuran, rt, 76%.

Synthesis of 1 (Scheme 4).

A similar coupling reaction of 15 with the aldehyde 14 gave an inseparable mixture of four diastereomers of the hydroxyl compound in 84% yield, which on oxidation with Dess-Martin periodinane provided 18 in quantitative yield (a:b = 1:1). The two diastereoisomers 18a and 18b were separated by chromatography. By similar ¹H NMR analysis and NOE experiments of 18, a 2.2% enhancement was observed in the signal for the 4-methyl group upon irradiation of the methylene proton of the methoxymethyloxy group in 18a, thereby confirming that the desired stereochemistry had been established. Deprotection of the *p*-methoxybenzyl group in the lactone 18a was accomplished using dichlorodicyanobenzoquinone in dichloromethane to give the β -hydroxy ketone 19a in 94% yield. The Jones oxidation of 19a under mild condition resulted in enol formation of the diketone 20a, the structure of which was elucidated by NMR.

An acid-catalyzed reaction of **20a** with boiling tetrahydrofuran in the presence of 3 *M* hydrochloric acid gave a mixture of diastereomers of **1** and its 2'-isomer **1b** (1:1 by NMR) in 70% yield. Such drastic conditions indeed would be inappropriate for the delicate ketone. Finally and much to our relief, deprotection of **20a** was achieved using bromotrimethylsilane in dichloromethane at -20° to give the spirolactone **1** as a single product in 98% yield. The synthetic **1** was found to be identical with the natural product by comparing the mp, $[\alpha]_D$, The advantages of this strategy are the short and efficient steps in one-pot direct construction of spirolactone by acidcatalysis *via* deprotection, cyclization and elimination.

EXPERIMENTAL

Melting points were measured with a Yanagimoto MP apparatus and are uncorrected. Specific rotations were measured with a JASCO Model DIP-370 polarimeter. Ir spectra were obtained on a JASCO A-102 infrared spectrophotometer. ¹H and ¹³C nmr spectra were recorded on a JEOL LA-300 (300 MHz) in deuterio chloroform solution, unless stated otherwise. Chemical shifts (ppm) are given downfield of tetramethylsilane. Mass spectra were determined on a JEOL AX-500 spectrometer. Elemental analyses were performed with a JMS AX-500 elemental analyzer by the staff at our Instrumental Measurement Center. Thin-layer chromatography (TLC) was performed with a glass plate coated Kieselgel 60 GF₂₅₄ (Merck). Column chromatography was carried out on silica gel 60 (Merck No. 7734; 63-200 µm). Solvents were dried (drying agent in parenthesis) and distilled prior to use: tetrahydrofuran and diethyl ether (sodium/benzophenone ketyl), benzene and dichloromethane (phosphorous pentoxide), and dimethylsulfoxide (calcium hydride). Organic solutions were dried over anhydrous sodium sulfate.

Ethyl 3-hydroxy-3-phenylpropanoate (5).

This compound was prepared from benzaldehyde and ethyl bromoacetate by the Reformatsky reaction in 85% yield as a colorless oil; bp $120-123^{\circ}/2$ mm Hg; (lit [4] $105^{\circ}/0.2$ mm Hg).

Ethyl 3-Phenyl-3-oxopropanoate (6).

This compound was prepared by the Collins oxidation [7] of **5** in 82% yield as a colorless oil: bp 115°/1 mm Hg; (lit [7] 95°/0.2 mm Hg); ir (film) 1725, 1670, 1620 cm⁻¹; ¹H nmr: δ 1.25 (t, 3H, J = 7.1 Hz), 3.99 (s, 2H), 4.21 (q, 2H, J = 7.1 Hz), 7.37-7.96 (m, 5H); ¹³C nmr: δ 14.1 (q), 46.0 (t), 61.5 (t), 126.0 (d), 128.5 (d), 128.8 (d), 133.7 (s), 167.5 (s), 192.5 (s). hrms: Calcd. for C₁₁H₁₂O₃: (M⁺) 192.0786. Found: m/z 192.0775.

Ethyl 3,3-Ethylenedioxy-3-phenylpropanoate (7).

A solution of **6** (10 mmoles), ethyleneglycol (20 mmoles) and *p*toluenesulfonic acid (20 mg) in benzene (50 ml) was refluxed for 15 hours. The cooled mixture was poured onto saturated aqueous sodium bicarbonate (50 ml), and the organic layer was washed with water and dried. The solvent was removed and the residue was distilled to give **7** in 73% yield as a colorless oil; bp 121°/2 mm Hg; ir (film) 1720 cm⁻¹; ¹H nmr: δ 1.15 (t, 3H, J = 7.1 Hz), 2.96 (s, 2H), 3.80-3.84 (m, 2H), 4.03-4.11 (m, 2H), 4.07 (q, 2H, J = 7.1 Hz), 7.27-7.52 (m, 5H); ¹³C nmr: δ 14.0 (q), 45.8 (t), 60.4 (t), 64.8 (t), 107.8 (s), 125.6 (d), 128.1 (d), 128.2 (d), 141.4 (s), 168.5 (s). hrms: Calcd. for C₁₃H₁₇O₄: (M + H)⁺ 237.1129. Found: m/z 237.1127.

Anal. Calcd for $C_{13}H_{16}O_4$: C, 66.09; H, 6.83. Found: C, 66.00; H, 6.88.

3,3-Ethylenedioxy-3-phenylpropanal (8).

To a solution of **7** (1.0 mmol) in toluene (10 ml) was added diisobutylaluminium hydride (1.5 *M* in toluene, 1.01 mmoles) at -78°. After stirring for 10 minutes the reaction mixture was quenched by addition of water. The precipitate was removed by filtration through celite. The filtrate was washed with water, dried and distilled to give **8** in 78% yield as white crystals; mp 87°; ir (nujol) 1710 cm⁻¹; ¹H nmr: δ 2.90 (d, 2H, J = 2.9 Hz), 3.78-3.90 (m, 2H), 4.03-4.14 (m, 2H), 7.29-7.50 (m, 5H), 9.76 (t, 1H, J = 2.9 Hz); ¹³C nmr: δ 52.7 (t), 64.7 (t), 108.2 (s), 125.3 (d), 128.5 (d x 2), 141.2 (s), 199.8 (d). MS (EI) m/z 192 (M⁺, 15%), 149 (90), 115 (100).

Anal. Calcd for C₁₁H₁₂O₃: C, 68.74; H, 6.29. Found: C, 68.48; H, 6.33.

(2*S*,3*S*)-3-Methyl-1,2-pentanediol (10).

To a stirred suspension of lithium aluminium hydride (4.36 g, 115 mmoles) in tetrahydrofuran (60 ml) was added a solution of (2*S*,3*S*)-2-acetoxy-3-methylvaleric acid **9** [5] (5.0 g, 28.7 mmoles) in tetrahydrofuran (30 ml) at 0°. After refluxing for 5 hours a small amount of water was added dropwise to decompose excess lithium aluminium hydride, and the precipitate was filtered off through celite and washed with ethyl acetate. The combined filtrates were dried and concentrated to give **10** as a colorless oil (2.8 g, 83%); bp 87°/9 mm Hg; $[\alpha]_D^{24}$ +5.9 (c 1.13, methanol); ir (film) 3400 cm⁻¹; ¹H nmr: δ 0.87 (d, 3H, J = 6.7 Hz), 0.91 (t, 3H, J = 7.3 Hz), 1.04-1.24 (m, 1H), 1.50-1.64 (m, 2H), 3.23 (br s, 2H), 3.49 (dd, 2H, J = 8.2 Hz, J = 17.1 Hz), 3.64-3.70 (m, 1H); ¹³C nmr: δ 11.2 (q), 14.6 (q), 25.0 (t), 37.6 (d), 64.2 (t), 76.0 (d).

(2S,3S)-3-Methyl-1,2-epoxypentane (11).

A mixture of **10** (2.5 g, 21.2 mmoles), benzyltriethylammonium chloride [6] (139 mg, 3 mol%) and 20% aqueous sodium hydroxide (25 ml) in 25 ml of dichloromethane was heated under reflux, and a solution of mesyl chloride (3.6g, 31.8 mmoles) in 40 ml of dichloromethane was added in portion. Heating and stirring was continued until complete consumption of the starting diol. The mixture was cooled to room temperature, and was poured into 25 ml of water. The organic layer was dried and distilled to give **11** (1.3 g, 62%) as a colorless liquid; bp 95°/760 mm Hg; $[\alpha]_D^{21}$ +3.8 (c 2.10, ethanol); ¹H nmr: δ 0.92 (d, 3H, J = 6.7 Hz), 0.96 (t, 3H, J = 7.3 Hz), 1.15-1.66 (m, 3H), 2.46-2.56 (m, 1H), 2.64-2.78 (m, 2H); ¹³C nmr: δ 11.2 (q), 15.1 (q), 27.2 (t), 37.6 (d), 45.6 (t), 56.9 (d).

(2*R*,3*S*)-2-(2-Hydroxy-3-methylpentyl)-1,3-dithiane (12).

To a solution of 1,3-dithiane (3.01 g, 25.1 mmoles) in tetrahydrofuran (40 ml) was added *n*-butyllithium (1.6 M in hexane, 14.2 ml, 22.9 mmols) at -40° under nitrogen and the mixture was stirred for 2 hours. A solution of 11 (2.09 g, 20.9 mmoles) in tetrahydrofuran (15 ml) was added to the reaction mixture, and stirred for additional 5 hours at -40°. After standing overnight at 0°, the reaction mixture was quenched with saturated aqueous ammonium chloride, and extracted with ether (4 x 100 ml). The combined organic layer was dried and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 5:1) to give **12** (3.63 g, 79%) as a pale yellow oil; $[\alpha]_{D}^{20}$ +28.3 (c 0.91, ethanol); ir (film) 3350 cm⁻¹; ¹H nmr: δ 0.89 (d, 3H, J = 6.8 Hz), 0.92 (t, 3H, J = 7.3 Hz), 1.12-1.21 (m, 1H), 1.44-1.53 (m, 2H), 1.80-1.96 (m, 3H), 2.02-2.19 (m, 1H), 2.84-2.94 (m, 4H), 3.78-3.90 (m, 1H), 4.28 (dd, 1H, J = 5.5 Hz, J = 8.3 Hz); ^{13}C nmr: δ 11.6 (q), 14.4 (q), 24.8 (t), 26.0 (t), 30.0 (t), 30.4 (t), 38.8 (d), 40.7 (t), 44.7 (d), 72.0 (d). MS (EI) m/z 220 (M⁺, 60%), 145 (100), 119 (75); hrms: Calcd. for C₁₀H₂₀OS₂: (M)⁺ 220.0956. Found: m/z 220.0982.

(2*R*,3*S*)-2-[2-(4-Methoxybenzyloxy)-3-methylpentyl]-1,3-dithiane (13).

To a suspension of sodium hydride (158 mg, 3.96 mmoles) in dimethylformamide (15 ml) was added 12 (581 mg, 2.64 mmoles) in dimethylformamide (3 ml) at room temperature, and the mixture was stirred for 1 hour. To the mixture was added p-methoxybenzyl chloride (1.07 ml, 7.90 mmoles), and the mixture was stirred for 21 hours at 50°. The reaction mixture was quenched with water, and extracted with ether (3 x 30 ml). The combined organic layer was dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 10/1) to give **13** (751 mg, 84%) as a colorless oil; $[\alpha]_D^{25}$ +43.0 (c 0.56, ethanol); ¹H nmr: δ 0.88 (d, 3H, J = 7.0 Hz), 0.92 (t, 3H, J = 7.3 Hz), 1.10-1.41 (m, 2H), 1.70-1.93 (m, 4H), 2.05-2.15 (m, 1H), 2.70-2.92 (m, 4H), 3.62-3.68 (m, 1H), 3.81 (s, 3H), 4.13 (dd, 1H, J = 4.2 Hz, J = 10.2 Hz), 4.39 (d, 1H, J = 11.0 Hz), 4.54 (d, 1H, J = 11.0 Hz), 6.88 (d, 2H, J = 8.8 Hz), 7.29 (d, 2H, J = 8.8 Hz); ¹³C nmr: δ 12.1 (q), 13.6 (q), 25.8 (t), 26.1 (t), 29.9 (t), 30.5 (t), 35.8 (t), 36.8 (d), 44.5 (d), 55.3 (q), 71.3 (t), 78.5 (d), 113.7 (d), 129.4 (d), 131.2 (s), 159.1 (s). MS (EI) m/z 340 (M⁺, 5%), 219 (74), 133 (100). hrms: Calcd. for C₁₈H₂₈O₂S₂: (M)⁺ 340.1531. Found: m/z 340.1509.

(3*R*,4*S*)-3-(4-Methoxybenzyloxy)-4-methylhexanal (14).

To a solution of **13** (1.10 g, 3.24 mmoles) in aqueous 80% acetonitrile (30 ml) was added mercury (II) chloride (2.19 g, 8.10 mmoles) and mercury (II) oxide (1.75 g, 8.10 mmoles) at room temperature. The mixture was heated at 60° with stirring for 2 hours. The cooled mixture was filtered through celite, and the filtrate was dried and concentrated. The residue was purified by chromatography (hexane/ethyl acetate, 10:1) to give **14** (680 mg, 84%) as a colorless oil; $[\alpha]_D^{23}$ +18.5 (c 0.73, ethanol); ir (film) 1724 cm⁻¹; ¹H nmr: δ 0.91 (d, 3H, J = 6.8 Hz), 0.94 (t, 3H, J = 7.3 Hz), 1.08-1.23 (m, 1H), 1.34-1.48 (m, 1H), 1.77-1.88 (m, 1H), 2.42 (ddd, 1H, J = 1.7 Hz, J = 3.5 Hz, J = 16.3 Hz), 2.60 (ddd, 1H, J = 2.7 Hz, J = 8.8 Hz, J = 16.3 Hz), 3.80 (s, 3H), 3.85-3.90 (m, 1H), 4.42 (d, 1H, J = 11.0 Hz), 4.52 (d, 1H, J = 11.0 Hz), 6.88 (d, 2H, J = 8.8 Hz), 7.24 (d, 2H, J = 8.8 Hz), 9.78 (dd, 1H, J = 1.8 Hz, J = 8.8 Hz). ¹³C nmr: δ 11.8 (q), 13.6 (q), 25.6 (t), 36.9 (d), 44.3 (t), 55.2 (q), 70.9 (t), 77.3 (d), 113.7 (d), 129.3 (d), 130.3 (s), 159.1 (s), 202.2 (d). hrms: Calcd. for C₁₅H₂₂O₃: (M)⁺ 250.1569. Found: m/z 250.1545.

(3S,4S) and (3R,4S)-3-[3',3'-(Ethylenedioxy)-3'-phenylpropanoyl]-3-methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone (**17a**) and (**17b**).

To a solution of lithium diisopropylamide, prepared from diisopropylamine (0.12 ml, 0.86 mmol) in tetrahydrofuran (10 ml) and *n*-butyllithium (1.6 *M* in hexane, 0.49 ml, 0.79 mmol), was added a solution of 15 (123 mg, 0.66 mmol) in tetrahydrofuran (5 ml) at -78° under nitrogen, and the mixture was stirred for 30 minutes. A solution of 8 (165 mg, 0.86 mmol) in tetrahydrofuran (5 ml) was added to the reaction mixture. After stirring for 3.5 hours at -78°, the mixture was quenched by addition of saturated aqueous ammonium chloride, and extracted with ether (4 x 10 ml). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 20:1) to give 16 (207 mg, 83%) as an inseparable mixture of four diastereomers; ir (film) 3500, 1765, 1620 cm⁻¹; To a solution of **16** (37 mg) in acetone (30 ml) was added Jones reagent (8 M, 0.04 ml) at 0°. After stirring for 1.5 hours, the mixture was quenched by addition of 2-propanol, followed by solid sodium bicarbonate. The mixture was filtered through celite and the filtrate was concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, 2:1) to give 17a (14 mg, 38%) and 17b (20 mg, 54%) as colorless needles, respectively. **17a**: mp 76°; $[\alpha]_D^{23}$ -29.3 (c 0.40, ethanol); ir (nujol) 1770, 1715 cm⁻¹; ¹H nmr: δ 1.28 (s, 3H), 3.17 (d, 1H, J = 17.2 Hz), 3.39 (s, 3H), 3.48 (d, 1H, J = 17.2 Hz), 3.73-3.85 (m, 2H), 4.00 (d, 1H, J = 8.8 Hz), 4.02-4.12 (m, 2H), 4.19 (d, 1H, J = 8.8 Hz, 4.97 (d, 1H, J = 6.0 Hz), 5.05 (d, 1H, J = 6.0 Hz), 5.12 (d, 1H, J = 6.0 Hz)1H, J = 17.0 Hz), 5.17 (d, 1H, J = 10.8 Hz), 5.58 (dd, 2H, J = $10.8 \text{ H$ 10.8 Hz, J = 17.4 Hz), 7.27-7.37 (m, 3H), 7.48-7.52 (m, 2H); ^{13}C nmr: δ 18.1 (q), 48.7 (t), 50.6 (s), 56.7 (q), 64.8 (t x 2), 73.6 (t), 88.8 (s), 94.6 (t), 108.3 (s), 117.7 (t), 125.8 (d), 128.1 (d), 128.2 (d), 135.7 (d), 141.9 (s), 171.6 (s), 203.0 (s). MS m/z 377 [(M + H)+, 5%], 317 (11), 271 (59), 185 (91), 149 (100), 93 (78). hrms: Calcd. for C₂₀H₂₅O₇: (M + H)⁺ 377.1600. Found: m/z 377.1579.

Anal. Calcd for C₂₀H₂₄O₇: C, 63.82; H, 6.43. Found: C, 63.75; H, 6.47.

Compound **17b** has mp 76°; $[\alpha]_D^{23}$ -30.2 (c 2.0, ethanol); ir (nujol) 1770, 1715 cm⁻¹; ¹H nmr: δ 1.00 (s, 3H), 3.17 (d, 1H, J = 17.3 Hz), 3.35 (s, 3H), 3.37 (d, 1H, J = 17.3 Hz), 3.74-3.82 (m, 2H), 3.83 (d, 1H, J = 8.5 Hz), 3.99-4.13 (m, 2H), 4.29 (d, 1H, J = 8.5 Hz), 4.79 (d, 1H, J = 6.4 Hz), 4.85 (d, 1H, J = 6.4 Hz), 5.17 (d, 1H, J = 17.6 Hz), 5.33 (d, 1H, J = 11.0 Hz), 6.15 (dd, 1H, J = 11.0 Hz, J = 17.6 Hz), 7.26-7.35 (m, 3H), 7.47-7.52 (m, 2H); ¹³C nmr: δ 17.7 (q), 48.7 (t), 49.3 (s), 56.5 (q), 64.7 (t), 64.9 (t), 74.6 (t), 87.3 (s), 94.6 (t), 108.1 (s), 117.7 (t), 125.8 (d), 128.0 (d), 128.1 (d), 135.2 (d), 141.9 (s), 170.2 (s), 200.7 (s). hrms: Calcd. for C₂₀H₂₅O₇: (M + H)⁺ 377.1600. Found: m/z 377.1583.

(-)-Hyperolactone C (3).

To a solution of 17a (9 mg, 0.024 mmol) in tetrahydrofuran (10 ml) was added a few drops of 3 *M* hydrochloric acid, and the mixture was heated at reflux for 11 hours. The mixture was

cooled and poured into aqueous sodium bicarbonate, and extracted with ether (3 x 20 ml). The combined extracts were dried and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 10:1) to give a crystalline **3** (6 mg, 93%); mp 106°; (lit [2] mp 104°); $[\alpha]_D^{25}$ -390 (c 0.018, ethanol) [lit [2] -356 (c 0.018, ethanol)]; ¹H nmr: δ 1.53 (s, 3H), 4.11 (d, 1H, J = 8.4 Hz), 4.97 (d, 1H, J = 8.4 Hz), 5.26 (d, 1H, J = 17.6 Hz), 5.27 (d, 1H, J = 10.8 Hz), 5.98 (s, 1H), 5.99 (dd, 1H, J = 10.8 Hz, J = 17.6 Hz), 7.49-7.54 (m, 2H), 7.58-7.63 (m, 1H), 7.83-7.86 (m, 2H); ¹³C nmr: δ 19.6 (q), 48.9 (s), 74.1 (t), 93.1 (s), 100.3 (d), 119.0 (t), 127.4 (d), 127.7 (s), 129.0 (d), 133.6 (d), 134.3 (d), 168.0 (s), 187.2 (s), 196.6 (s). MS (EI) m/z 270 (M⁺, 35%), 252 (8), 225 (15), 211 (23), 197 (4), 187 (100), 173 (11), 147 (15), 105 (5), 129 (9), 102 (56). hrms: Calcd. for C₁₆H₁₄O₄: (M)⁺ 270.0892. Found: m/z 270.0909.

5-Epi-hyperolactone C (3a).

The similar reaction of the β -methoxymethyloxy isomer **17b** (41 mg) as in the synthesis of **3** gave, after chromatography, **3a** (29.4 mg, 100%) as a wax; $[\alpha]_D^{20} + 283$ (c 0.054, ethanol); ¹H nmr: δ 1.30 (s, 3H), 4.38 (d, 1H, J = 8.6 Hz), 4.78 (d, 1H, J = 8.6 Hz), 5.34 (d, 1H, J = 17.4 Hz), 5.37 (d, 1H, J = 10.8 Hz), 6.04 (s, 1H), 6.06 (dd, 1H, J = 10.8 Hz, J = 17.4 Hz), 7.48-7.54 (m, 2H), 7.58-7.64 (m, 1H), 7.83-7.86 (m, 2H); ¹³C nmr: δ 15.4 (q), 48.9 (s), 73.3 (t), 92.3 (s), 100.4 (d), 116.2 (t), 127.4 (s), 127.5 (d), 129.0 (d), 133.6 (d), 136.8 (d), 168.1 (s), 187.4 (s), 196.6 (s). MS (EI) m/z 270 (M⁺, 46%), 252 (8), 211 (25), 187 (100), 173 (16), 147 (14), 129 (9), 102 (58). hrms: Calcd. for C₁₆H₁₄O₄: (M)⁺ 270.0892. Found: m/z 270.0880.

(3S,4S, 3'R, 4'S) and (3R,4S,3'R,4'S)-3-[3'-(4-Methoxybenzyloxy)-4'-methylhexanoyl]-3-methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone (**18a**) and (**18b**).

The procedure for 16 was employed with 14 and 15 gave an alcohol as an inseparable mixture of four diastereomers in 84% yield. To a solution of the resulting alcohol (269 mg, 0.62 mmol) in dichloromethane (20 ml) was added Dess-Martin periodinane (915 mg, 2.16 mmoles), and the mixture was stirred for 2 hours at room temperature. The reaction was quenched by addition of 2-propanol, and the precipitate was filtered off. The filtrate was washed with 10% sodium thiosulfate solution, and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate, 20:1) to give 18a (134 mg, 50%) and 18b (134 mg, 50%). 18a (colorless oil); $[\alpha]_D^{21}$ -0.20 (c 2.7, ethanol); ¹H nmr: δ 0.89 (d, 3H, J = 6.8 Hz), 0.91 (t, 3H, J = 7.3 Hz), 1.03-1.18 (m, 1H), 1.30 (s, 3H), 1.23-1.40 (m, 1H), 1.68-1.80 (m, 1H), 2.59 (dd, 1H, J = 2.7 Hz, J = 19.0 Hz), 2.99 (dd, 1H, J = 9.0 Hz, J = 19.0 Hz), 3.43 (s, 3H), 3.79 (s, 3H), 3.94-3.97 (m, 1H), 4.00 (d, 1H, J = 8.6 Hz), 4.18 (d, 1H, J = 8.6 Hz), 4.41 (d, 1H, J = 10.8 Hz), 4.48 (d, 1H, J = 10.8 Hz), 4.99 (s, 2H), 5.11 (d, 1H, J = 10.8 Hz), 5.14 (d, 1H, J = 17.4 Hz), 5.75 (dd, 1H, J = 10.8 Hz, J = 17.4 Hz), 6.84 (d, 2H, J = 8.4 Hz), 7.22 (d, 2H, J = 8.4 Hz); ${}^{13}C$ nmr: δ 12.0 (q), 13.9 (q), 17.7 (q), 25.6 (t), 37.1 (d), 41.6 (t), 50.7 (s), 55.2 (q), 56.6 (q), 71.6 (t), 73.9 (t), 77.3 (d), 88.8 (s), 94.5 (t), 113.5 (d), 117.5 (t), 129.2 (d), 131.1 (s), 135.7 (d), 158.9 (s), 171.8 (s), 206.4 (s). hrms: Calcd. for C24H34O7: (M)+ 434.2304. Found: m/z 434.2328. 18b (colorless oil); $[\alpha]_D^{22}$ -11.3 (c 3.76, ethanol); ¹H nmr: δ 0.88 (d, 3H, J = 6.8 Hz), 0.91 (t, 3H, J = 7.3 Hz), 1.11 (s, 3H), 1.23-1.41 (m, 2H), 1.70- $1.78 \text{ (m, 1H)}, 2.31 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 2.8 \text{ Hz}, \text{J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 1H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H, J} = 18.1 \text{ Hz}), 3.00 \text{ (dd, 2H,$ J = 9.4 Hz, J = 18.1 Hz), 3.35 (s, 3H), 3.78 (s, 3H), 3.88 (d, 1H, J = 8.4 Hz), 3.96-4.03 (m, 1H), 4.36 (d, 1H, J = 8.4 Hz), 4.41 (d, 1H, J = 10.6 Hz), 4.45 (d, 1H, J = 10.6 Hz), 4.77 (d, 1H, J = 6.4 Hz), 4.81 (d, 1H, J = 6.4 Hz), 5.17 (d, 1H, J = 17.8 Hz), 5.34 (d, 1H, J = 11.0 Hz), 6.23 (dd, 1H, J = 11.0 Hz, J = 17.8 Hz), 6.84 (d, 2H, J = 8.6 Hz), 7.22 (d, 2H, J = 8.6 Hz); ¹³C nmr: δ 12.0 (q), 13.9 (q), 18.3 (q), 26.0 (t), 37.0 (d), 41.4 (t), 49.5 (s), 55.2 (q), 56.5 (q), 72.0 (t), 74.9 (t), 77.9 (d), 87.2 (s), 94.7 (t), 113.6 (d), 117.6 (t), 129.5 (d), 130.8 (s), 135.1 (d), 159.0 (s), 170.7 (s), 204.8 (s). hrms: Calcd. for C₂₄H₃₄O₇: (M)⁺ 434.2304. Found: m/z 434.2281.

(3S,4S,3'R,4'S) and (3R,4S,3'R,4'S)-3-(3'-Hydroxy-4'-methyl-hexanoyl)-3-methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone (**19a**) and (**19b**).

A mixture of 18a (131 mg, 0.3 mmol), dichlorodicyanobenzoquinone (DDQ) (177 mg, 0.76 mmol) in dichloromethane (9 ml) and water (0.5 ml) was stirred at room temperature for 90 minutes. The mixture was diluted with saturated aqueous sodium bicarbonate solution and extracted with dichloromethane (3 x 30 ml). The organic layer was dried and concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, 10:1) to give 19a (84 mg, 94% yield) as a colorless oil; $[\alpha]_D^{23}$ +1.87 (c 1.65, ethanol); ir (film) 3545, 1790, 1710 cm⁻¹; ¹H nmr: δ 0.87 (d, 3H, J = 6.6 Hz), 0.90 (t, 3H, J = 7.1 Hz), 1.04-1,26 (m, 1H), 1.34 (s, 3H), 1.43-1.57 (m, 2H), 2.74 (d, 1H, J = 9.5 Hz), 2.79 (dd, 1H, J = 4.2 Hz, J = 15.0 Hz), 3.43 (s, 3H), 3.88-3.95 (m, 1H), 4.12 (d, 1H, J = 8.8Hz), 4.25 (d, 1H, J = 8.8 Hz), 4.96 (s, 2H), 5.23 (d, 1H, J = 17.4 Hz), 5.29 (d, 1H, J = 10.8 Hz), 5.78 (dd, 1H, J = 10.8 Hz, J = 17.4 Hz); ¹³C nmr: δ 11.4 (q), 14.4 (q), 17.0 (q), 24.8 (t), 39.7 (d), 44.3 (t), 50.6 (s), 56.6 (q), 70.8 (d), 74.0 (t), 87.7 (s), 94.5 (t), 117.7 (t), 135.9 (d), 171.5 (s), 208.8 (s). hrms (FAB): Calcd. for C₁₆H₂₆O₆Na⁺: (M + Na)⁺ 337.1627. Found: m/z 337.1640.

The β -methoxymethyloxy isomer **18b** was treated as in the synthesis of **19a** to give a colorless oil **19b** (94% yield); [α]₂²³-11.1 (c 2.55, ethanol); ¹H nmr: δ 0.87 (d, 3H, J = 6.6 Hz), 0.90 (t, 3H, J = 7.3 Hz), 1.14 (s, 3H), 1.10-1.17 (m, 1H), 1.41-1.56 (m, 2H), 2.63-2.77 (m, 2H), 2.92 (d, 1H, J = 3.3 Hz, OH), 3.42 (s, 3H), 3.90 (d, 1H, J = 8.6 Hz), 3.90-3.96 (m, 1H), 4.36 (d, 1H, J = 8.6 Hz), 4.86 (s, 2H), 5.24 (d, 1H, J = 17.6 Hz); 5.38 (d, 1H, J = 10.8 Hz), 6.25 (dd, 1H, J = 10.8 Hz, J = 17.6 Hz); ¹³C nmr: δ 11.4 (q), 14.4 (q), 18.2 (q), 24.8 (t), 39.5 (d), 43.6 (t), 49.3 (s), 56.6 (q), 70.6 (d), 74.8 (t), 87.1 (s), 94.6 (t), 118.0 (t), 134.9 (d), 170.2 (s), 207.3 (s). hrms (FAB): Calcd. for C₁₆H₂₆O₆Na⁺: (M + Na)⁺ 337.1627. Found: m/z 337.1640.

(3S,4S,4'S) and (3R, 4S, 4'S)-3-(3'-Oxo-4'-methylhexanoyl)-3-methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone (**20a**) and (**20b**).

To a solution of **19a** (28.1 mg, 0.09 mmol) in acetone (5 ml) was added an excess amount of Jones reagent (8 *M* solution) at -20°. After stirring for 2 hours, the reaction mixture was quenched by addition of 2-propanol and solid sodium bicarbonate. The mixture was filtered through Celite, and the filtrate was concentrated *in vacuo*. The residue was purified by chromatography (hexane/ethyl acetate, 9:1) to give **20a** (18.3 mg, 66% yield) as a pale yellow oil; $[\alpha]_D^{21}$ -11.7 (c 0.63, ethanol); ir (film) 2970, 1790, 1600 cm⁻¹; ¹H nmr: δ 0.89 (t, 3H, J = 7.5 Hz), 1.14 (d, 3H, J = 7.0 Hz), 1.36 (s, 3H), 1.42-1.54 (m, 1H), 1.59-1.73 (m, 1H), 2.22-2.34 (m, 1H), 3.47 (s, 3H), 4.12 (d, 1H, J = 8.6 Hz), 4.33 (d, 1H, J = 8.6 Hz), 4.94 (d, 1H, J = 5.9 Hz), 5.09 (d, 1H, J = 5.9 Hz), 5.13 (d, 1H, J = 17.4 Hz), 5.16 (d, 1H, J = 11.0 Hz), 5.73 (dd, 1H, J = 11.0 Hz, J = 17.4 Hz), 5.92 (s, 1H); ¹³C nmr: δ 11.6

(q), 17.2 (q), 17.9 (q), 27.1 (t), 43.6 (d), 50.7 (s), 56.8 (q), 73.9 (t), 85.8 (s), 94.4 (t), 97.8 (d), 116.3 (t), 136.4 (d), 172.2 (s), 190.8 (s), 196.5 (s). hrms: Calcd. for $C_{16}H_{24}O_6$: (M)⁺ 312.1573. Found: m/z 312.1563.

The β -methoxymethyloxy isomer **19b** was treated as in the synthesis of **20a** to give pale yellow crystals **20b** (65%); mp 54.5 $^{\rm o}$ C; $[\alpha]_{\rm D}^{20}$ +13.6 (c 0.89, ethanol); $^{\rm 1}$ H nmr: δ 0.90 (t, 3H, J = 7.5 Hz), 1.15 (d, 3H, J = 6.8 Hz), 1.16 (s, 3H), 1.41-1.55 (m, 1H), 1.59-1.73 (m, 1H), 2.25-2.37 (m, 1H), 3.43 (s, 3H), 3.99 (d, 1H, J = 8.4 Hz), 4.43 (d, 1H, J = 8.4 Hz), 4.83 (d, 1H, J = 6.2 Hz), 4.92 (d, 1H, J = 6.2 Hz), 5.17 (d, 1H, J = 17.6 Hz), 5.32 (d, 1H, J = 11.0 Hz), 5.95 (s, 1H), 6.17 (dd, 1H, J = 11.0 Hz, J = 17.6 Hz); $^{\rm 13C}$ nmr: δ 11.6 (q), 17.0 (q), 18.9 (q), 27.1 (t), 43.8 (d), 49.9 (s), 56.7 (q), 74.8 (t), 84.9 (s), 94.4 (t), 98.2 (d), 116.9 (t), 135.6 (d), 171.2 (s), 188.2 (s), 197.9 (s).

(2'S,5S,9S)-Isomer (1): (-)-Hyperolactone A.

To a cooled solution of 20a (7.5 mg) in dichloromethane (5 ml) was added trimethylsilyl bromide (0.03 ml, 0.22 mmol) at -20°, and the mixture was stirred for 14 hours. The reaction was quenched by addition of saturated aqueous sodium bicarbonate and the mixture was extracted with dichloromethane (3 x 10 ml). The combined extracts were washed with water, dried and concentrated. The residue was chromatographed on silica gel (hexane/ethyl acetate, 10:1) to give 1 (5.9 mg, 98%) as a single product; mp 57° (recrystallized from ether and n-hexane); [lit [1] mp 57° for natural hyperolactone A]; $[\alpha]_D^{24}$ -265 (c 0.13, methanol); [lit [1] $[\alpha]_D^{24}$ -229 (c 0.13, methanol)]; ¹H nmr: δ 0.96 (t, 3H, J = 7.5 Hz), 1.25 (d, 3H, J = 6.8 Hz), 1.41 (s, 3H), 1.59-1.78 (m, 2H), 2.63-2.75 (m, 1H), 4.05 (d, 1H, J = 8.4 Hz), 4.90 (d, 1H, J = 8.4 Hz), 5.24 (d, 1H, J = 17.4Hz), 5.28 (d, 1H, J = 10.8 Hz), 5.38 (s, 1H), 5.94 (dd, 1H, J = 10.8 Hz, J = 17.4 Hz). ¹³C nmr: δ 11.0 (q), 16.9 (q), 19.3 (q), 27.1 (t), 37.1 (d), 48.4 (s), 74.0 (t), 92.5 (s), 102.1 (d), 118.8 (t), 134.3 (d), 168.0 (s), 197.3 (s), 200.2 (s). MS (EI) m/z 250 (M+, 31%), 193 (53), 177 (42), 167 (100). hrms: Calcd. for C₁₄H₁₈O₄: (M)⁺ 250.1205. Found: m/z 250.1216.

(2'S,5R,9S)-Isomer (1a).

The β -methoxymethyloxy isomer **20b** was treated as in the synthesis of **1** to give a colorless was **1a** (89% yield) as a single product; $[\alpha]_D^{21}$ +194 (c 0.20, ethanol); ¹H nmr: δ 0.99 (t, 3H, J = 7.3 Hz), 1.23 (s, 3H), 1.27 (d, 3H, J = 7.0 Hz), 1.54-1.68 (m, 1H), 1.67-1.79 (m, 1H), 2.61-2.73 (m, 1H), 4.31 (d, 1H, J = 8.6 Hz), 4.71 (d, 1H, J = 8.6 Hz), 5.27 (d, 1H, J = 17.4 Hz), 5.30 (d, 1H, J = 10.8 Hz), 5.43 (s, 1H), 5.93 (dd, 1H, J = 10.8 Hz, J = 17.4 Hz); ¹³C nmr: δ 11.3 (q), 15.4 (q), 17.2 (q), 27.1 (t), 37.3 (d), 48.4 (s), 73.2 (t), 90.3 (s), 102.2 (d), 116.0 (t), 136.8 (d), 168.0 (s), 197.4 (s), 200.5 (s). MS (EI) m/z 250 (M⁺, 23%), 193 (42), 177 (47), 167 (100). hrms: Calcd. for C₁₄H₁₈O₄: (M)⁺ 250.1205. Found: m/z 250.1215.

(2'*R*,5*S*,9*S*)-Isomer (1b).

The α -methoxymethyloxy isomer **20a** was treated with boiling tetrahydrofuran containing 3 *M* hydrochloric acid to give a mixture of **1** and its 2'-isomer **1b** (88% yield), which was separated by HPLC [Shimadzu LC-3A; column: Cosmosil 5SI; *n*-hexanediethyl ether (95:5); 5 ml/minute]. The first peak (Rt 21.25 minutes) gave **1** and second peak (Rt 21.75 minutes) gave **1b** as a colorless wax. **1b**: ¹H nmr: δ 0.97 (t, 3H, J = 7.3 Hz), 1.25 (d, 3H, J = 7.3 Hz), 1.41 (s, 3H), 1.50-1.62 (m, 1H), 1.62-1.80 (m, 1H), 2.69 (sextet, 1H, J = 7.3 Hz), 4.04 (d, 1H, J = 8.5 Hz), 4.89

(d, 1H, J = 8.5 Hz), 5.24 (d, 1H, J = 17.1 Hz), 5.28 (d, 1H, J = 11.0 Hz), 5.37 (s, 1H), 5.93 (dd, 1H, J = 11.0 Hz, J = 17.1 Hz); ¹³C nmr: δ 11.3 (q), 17.3 (q), 19.3 (q), 27.2 (t), 37.3 (d), 48.4 (s), 74.0 (t), 92.5 (s), 102.2 (d), 118.9 (t), 134.4 (d), 168.0 (s), 197.4 (s), 200.2 (s). hrms: Calcd. for C₁₄H₁₈O₄: (M)⁺ 250.1205. Found: m/z 250.1195.

Preparation of the Key Intermediate 15.

(2*S*,3*R*)-5-Benzyloxy-3-ethoxycarbonyl-2-hydroxy-3-methylpentanoic acid (**21b**).

To a solution of lithium diisopropylamide, prepared from diisopropylamine (11.3 ml, 80.4 mmoles) and n-butyllithium (1.6 M in hexane, 45.6 ml, 73.7 mmoles) in tetrahydrofuran (200 ml), was added a solution of diethyl (2S,3R)-(+)-2-hydroxy-3-methylsuccinate [8] (6.8 g, 33.4 mmoles) in tetrahydrofuran (12 ml) at -78° and the mixture was stirred for 3 hours. To the cooled (-78) solution was added benzyl 2-iodoethyl ether [9] (13.1 g, 50 mmoles) within 5 minutes and the mixture was stirred for 8 hours, then at 0° for 13 hours. After quenching by addition of a solution of acetic acid (8 ml) in 15 ml of ether, the mixture was poured into water and extracted with ether (4 x 100 ml). The combined extracts were washed with aqueous sodium bicarbonate, dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 10:1) to give 21a (12.46 g, 67%) as a pale yellow oil; $[\alpha]_D^{27}$ +12.9 (c 0.70, ethanol); ir (film) 3490, 1715 cm⁻¹; ¹H nmr: δ 1.21 (t, 3H, J = 7.1 Hz), 1.22 (s, 3H), 1.27 (t, 3H, J = 7.1 Hz), 1.80-1.99 (m, 1H), 2.10-2.22 (m,1H), 3.55 (t, 2H, J = 6.5 Hz), 3.65 (d, 1H, J = 9.8 Hz, OH), 4.12 (q, 2H, J = 7.1 Hz), 4.24 (q, 2H, J = 7.1 Hz), 4.32 (d, 1H, J = 9.8 Hz), 4.47 (s, 2H), 7.27-7.33 (m, 5H); MS (EI) m/z 338 (M+, 15%), 293 (4), 231 (16), 204 (23), 157 (35), 131 (30), 91 (100); hrms: Calcd. for C₁₈H₂₆O₆: (M)⁺ 338.1729. Found: m/z 338.1728. To the resulting diester 21a (5.11 g) in methanol (40 ml) was added a solution of potassium hydroxide (2.85 g, 3.4 equivalents) in methanol (30 ml) and water (2 ml) at -40°, and the mixture was stirred for 1 hour, then at room temperature for 3 days. The mixture was acidified by addition of 3 *M* hydrochloric acid to pH-2 and extracted with ether (4 x 100 ml). The combined extracts were dried and concentrated in vacuo to give **21b** (4.55 g, 97%) as a pale yellow syrup; $[\alpha]_D^{27}$ +8.4 (c 0.71, ethanol); ir (film) 3400, 1700 cm⁻¹; ¹H nmr: δ 1.21 (t, 3H, J = 7.1 Hz), 1.25 (s, 3H), 1.99-2.15 (m, 2H), 3.57-3.66 (m, 2H), 4.13 (q, 2H, J = 7.1 Hz), 4.51 (s, 2H), 7.27-7.33 (m, 5H); ¹³C nmr: δ 13.9 (q), 18.6 (q), 34.8 (t), 49.0 (s), 60.9 (t), 61.4 (t), 66.4 (t), 73.4 (t), 74.4 (d), 128.0 (d), 128.5 (d x 2), 137.0 (s), 174.4 (s), 174.9 (s). hrms: Calcd. for C₁₆H₂₂O₆: (M)⁺ 310.1416. Found: m/z 310.1429.

(3*S*,4*S*)-4-(2-Benzyloxyethyl)-3-hydroxy-4-methyl-γ-butyrolactone (**22a**).

To a solution of **21b** (2.29 g, 7.4 mmoles) in tetrahydrofuran (100 ml) was added lithium triethylborohydride (1.0 *M* in tetrahydrofuran, 44.3 ml, 44.3 mmoles) at -78°, and the mixture was stirred for 3 hours, then at room temperature for 40 hours. To the reaction mixture was added 17 ml of water at -10°, and the aqueous layer was extracted with ether (3 x 50 ml). The combined extracts were dried and concentrated *in vacuo*. The residue was purified by column chromatography (hexane/ethyl acetate, 2:1) to give **22a** (1.37 g, 74%) as a colorless oil; $[\alpha]_D^{27}$ +6.0 (c 0.80, ethanol); ir (film) 3400, 1760, 1620 cm⁻¹; ¹H nmr: δ 1.26 (s, 3H), 1.58-1.94 (m, 2H), 3.48 (dt, 1H, J = 2.4 Hz, J = 10.1 Hz), 3.57-3.66 (m, 1H), 3.97 (d, 1H, J = 11.7 Hz), 4.59 (d, 1H, J = 11.7 Hz), 7.30-7.38 (m, 5H); ¹³C nmr: δ

23.5 (q), 33.8 (t), 42.5 (s), 66.0 (t), 73.1 (t), 74.7 (t), 75.4 (d), 127.9 (d), 128.1 (d), 128.4 (d), 136.7 (s), 176.5 (s). hrms: Calcd. for $C_{14}H_{18}O_4$: (M)⁺ 250.1205. Found: m/z 250.1219.

(3S,4S)-4-(2-Benzyloxyethyl)-3-methoxymethyloxy-4-methyl- γ -butyrolactone (**22b**).

To a solution of **22a** (730 mg, 2.92 mmoles) and diisopropylethylamine (5.1 ml, 29.0 mmoles) in dichloromethane (50 ml) was added chloromethoxymethane (1.11 ml, 14.6 mmoles), and the mixture was heated at reflux for 48 hours. The mixture was cooled, diluted with water, and extracted with dichloromethane (4 x 20 ml). The combined extracts were dried and concentrated *in vacuo*. The residue was purified by column chromatography (hexane/ethyl acetate, 3:1) to give **22b** (795 mg, 93%) as a pale yellow oil; $[\alpha]_{27}^{27}$ -47.7 (c 2.3, ethanol); ir (film) 1770 cm⁻¹; ¹H nmr: δ 1.19 (s, 3H), 1.72-1.95 (m, 2H), 3.41 (s, 3H), 3.57-3.62 (m, 2H), 3.88 (d, 1H, J = 9.0 Hz), 3.99 (s, 1H), 4.39 (d, 1H, J = 9.0 Hz), 4.47 (s, 2H), 4.69 (d, 1H, J = 6.8 Hz), 4.99 (d, 1H, J = 6.8 Hz), 7.28-7.35 (m, 5H); ¹³C nmr: δ 21.1 (q), 31.3 (t), 42.1 (s), 56.0 (q), 66.4 (t), 73.2 (t), 75.1 (t), 78.3 (d), 95.9 (t), 127.6 (d x 2), 128.4 (d), 138.1 (s), 175.0 (s). MS (EI): m/z 249 ([M-CH₂OCH₃]⁺, 40%), 143 (65), 91 (100).

(3S,4S)-4-(2-Hydroxyethyl)-3-methoxymethyloxy-4-methyl- γ -butyrolactone (**23a**).

A mixture of **22b** (622 mg, 2.11 mmoles) and 10% palladium on carbon (300 mg) in ethanol (25 ml) was vigorously stirred under an atmosphere of hydrogen at room temperature for 44 hours. The mixture was filtered and the filtrate was concentrated *in vacuo*. The residue was purified by column chromatography (hexane/ethyl acetate, 1:1) to give **23a** (414 mg, 96%) as a colorless oil; $[\alpha]_D^{22}$ -94.6 (c 0.65, ethanol); ir (film) 3425, 1760 cm⁻¹; ¹H nmr: δ 1.24 (s, 3H), 1.79 (t, 2H, J = 6.6 Hz), 2.21 (br s, 1H, OH), 3.44 (s, 3H), 3.70-3.82 (m, 2H), 3.93 (d, 1H, J = 9.0 Hz), 4.05 (s, 1H), 4.35 (d, 1H, J = 9.0 Hz), 4.73 (d, 1H, J = 6.8 Hz), 5.01 (d, 1H, J = 6.8 Hz); ¹³C nmr: δ 21.1 (q), 34.6 (t), 42.1 (s), 56.0 (q), 58.7 (t), 75.0 (t), 78.3 (d), 96.0 (t), 175.0 (s). MS (EI) m/z ([M-OCH₂OCH₃]⁺, 16%), 129 (11), 113 (43), 99 (100), 91 (11), 85 (27). hrms: Calcd. for C₇H₁₁O₃: (M - OCH₂OCH₃)⁺ 143.0708. Found: m/z 143.0722.

(3*S*,4*S*)-3-Methoxymethyloxy-4-methyl-4-[2-(2-nitrophenyl)-seleno]-γ-butyrolactone (**23b**).

A mixture of 23a (673 mg, 3.3 mmoles), 2-nitrophenylselenocyanate (1.05 g, 4.62 mmoles) and *n*-tributylphosphine (1.15 ml, 4.62 mmoles) in 30 ml of tetrahydrofuran was stirred at room temperature for 1 hour. The reaction mixture was diluted with water and extracted with ether (5 x 20 ml). The combined extracts were dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 4/1) to give **23b** (1.13 g, 88%) as yellow crystals; mp 74-75°; $[\alpha]_D^{27}$ -37.0 (c 0.24, ethanol); ir (nujol) 1770, 1590, 1560, 1510 cm⁻¹; ¹H nmr: δ 1.29 (s, 3H), 1.87-2.06 (m, 2H), 2.85-3.00 (m, 2H), 3.46 (s, 3H), 4.01 (d, 1H, J = 9.4 Hz), 4.18 (s, 1H), 4.26 (d, 1H, J = 9.4 Hz), 4.78 (d, 1H, J = 6.8 Hz), 5.08 (d, 1H, J = 6.8 Hz), 7.30-7.35 (m, 1H), 7.50-7.59 (m, 3H), 8.28 (dd, 1H, J = 1.3 Hz, J = 8.3 Hz). ¹³C nmr: δ 20.4 (t), 22.2 (q), 33.3 (t), 44.7 (s), 56.2 (q), 74.3 (t), 77.5 (d), 96.2 (t), 125.5 (d), 126.5 (d), 128.9 (d), 132.8 (d), 133.7 (s), 146.7 (s), 174.8 (s). hrms: Calcd. for C₁₅H₁₉NO₆Se: (M)⁺ 389.0377. Found: m/z 389.0399.

Anal. Calcd for C₁₅H₁₉NO₆Se: C, 46.40; H, 4.93; N, 3.61. Found: C, 46.45; H, 4.98; N, 3.61. (3S,4S)-3-Methoxymethyloxy-4-methyl-4-vinyl- γ -butyrolactone (15).

To a solution of 23b (1.11 g, 2.85 mmoles) in tetrahydrofuran (40 ml) was added dropwise 30% hydrogen peroxide (1.58 ml, 17.2 mmoles) at 0°, and the mixture was warmed to room temperature. After stirring for 5 hours, the reaction mixture was diluted with water and extracted with ether (4 x 50 ml). The combined extracts were washed with sodium thiosulfate solution, dried and concentrated in vacuo. The residue was purified by column chromatography (hexane/ethyl acetate, 6:1) to give 15 (403 mg, 76%) as an oil; $[\alpha]_D^{25}$ -64.7 (c 2.20, ethanol); ir (film) 1780, 1630 cm⁻¹; ¹H nmr: δ 1.32 (s, 3H), 3.43 (s, 3H), 3.98 (d, 1H, J = 9.0 Hz), 4.13 (s, 1H), 4.28 (d, 1H, J = 9.0 Hz), 4.71 (d, 1H, J = 6.8 Hz), 4.98 (d, 1H, J = 7.0 Hz), 5.23 (d, 1H, J = 17.6 Hz), 5.27 (d, 1H, J = 10.8 Hz), 5.97 (dd, 1H, J = 10.8 Hz, J = 17.5 Hz); ¹³C nmr: δ 20.4 (q), 45.6 (s), 55.9 (q), 74.1 (t), 78.4 (d), 96.0 (t), 115.9 (t), 136.3 (d), 174.2 (s). hrms (CI): Calcd. for C₉H₁₅O₄: (M + H)⁺ 187.0934. Found: m/z 187.0954.

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